

Lab Report 2: Characterization of Synthesis of Cadmium

Name: Elia Kufner

Matriculation No.: 21

Course: Chemie 1

Reaction:

Effect of persulfate on the stability of cadmium

Time (h)	Cd^{2+} (ppm)	$\text{S}_2\text{O}_8^{2-}$ (ppm)
0	1000	1000
1	1000	1000
2	1000	1000
3	1000	1000
Average Cd^{2+} (ppm)		1000
Average $\text{S}_2\text{O}_8^{2-}$ (ppm)		1000

Reaction: Pb^{2+} complex: $\text{Pb}(\text{CH}_3\text{COO})_2$

a) Characterization of lead:

Stock solution: 1 mg/L Pb^{2+} in 10 mL of 1 M HNO_3 (pH 1.0)

b) Cadmium complex: $\text{Pb}(\text{CH}_3\text{COO})_2$

Stock solution: 1 mg/L Pb^{2+} in 10 mL of 1 M HNO_3 (pH 1.0)

Reaction: $\text{Pb}(\text{CH}_3\text{COO})_2$

Reaction of Pb^{2+} with $\text{S}_2\text{O}_8^{2-}$ in 1 M HNO_3 (pH 1.0)

Reaction of Pb^{2+} with $\text{S}_2\text{O}_8^{2-}$ in 1 M HNO_3 (pH 1.0)

Reaction: Pb^{2+} complex: $\text{Pb}(\text{CH}_3\text{COO})_2$ (see $\text{Pb}(\text{CH}_3\text{COO})_2$)
Reaction of Pb^{2+} with $\text{S}_2\text{O}_8^{2-}$ in 1 M HNO_3 (pH 1.0)

Reaction of Pb^{2+} with $\text{S}_2\text{O}_8^{2-}$ in 1 M HNO_3 (pH 1.0)
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Reaction: Pb^{2+} complex: $\text{Pb}(\text{CH}_3\text{COO})_2$ (see $\text{Pb}(\text{CH}_3\text{COO})_2$)
Reaction of Pb^{2+} with $\text{S}_2\text{O}_8^{2-}$ in 1 M HNO_3 (pH 1.0)

Gravimetric Determination Of Calcium Lab Report Answers

Terry C. Jones



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