

STOICHIOMETRY QUIZ

4 K 11 T

Assessment of Learning (Product)

These questions are based on the following unbalanced chemical equation.



1a. Write the balanced chemical equation? (2K)

1b. If you were given 1 mol of C_6H_{14} and 10 mole of O_2 , which reactant is limiting? (2K)

1c. Calculate the mass of oxygen needed to produce 4.5×10^{24} molecules of carbon dioxide. (4T)

2a. If 1 g of C_6H_{14} is combined with 5 g of oxygen, what is the mass of water vapour produced? (4T)

2b. When Dr. Mathewson did this reaction in question 2a he obtained 0.45 g of water. What was the percentage yield? (3T)

Grade 11 Stoichiometry Questions And Memo

Sabine Zange



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