Ionic Chemical Formulas/PreAICE Chemistry

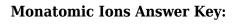
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Directions: Using the Chemical Formulas program sheet, fill in the table below. Use subscripts & superscripts. Remember to use roman numerals to show the valence of the transition metals in the compound name.

METAL CATION (+)	FORMULA	NONMETAL ANION (-)	FORMULA	FORMULA	COMPOUND NAME				
sodium	Na+1	hydroxide	OH-	NaOH	sodium hydroxide				
magnesium	Mg+2	phosphate	PO4-3	Mg3(PO4)2	magnesium phosphate				
calcium	Ca+2	chloride	CI- SO3-2 OH-	CaCl2	calcium chloride gold(I) sulfite copper(II) hydroxide				
gold(I)	Au+1	sulfite		Au2SO3					
copper(II)	Cu+2	hydroxide		Cu(OH)2					
Iron(III)	Fe+3	telluride	Te-2	Fe2Te3	iron(III) telluride				
magnesium	Mg+2	bromide	Br-1	MgBr2	magnesium bromide				
manganese(II)	Mn+2	sulfide	S-2	MnS	manganese(II) sulfide				
boron	B+3	oxide	0-2	B203	boron oxide				
potassium	K+	nitrate	NO3-	KN03	potassium nitrate				
barium	Ba+2	bromide	Br-	BaBr2	barium bromide				
strontium	Sr+2	sulfate	SO4-2	SrSO4	strontium sulfate				
iron(III)	Fe+3	oxide	0-2	Fe2O3	iron(III) oxide				
silver(I)	Ag+1	chloride	CI-	AgCl	silver(I) chloride				
aluminum	Al+3	carbonate	CO3-2	AI2(CO3)3	aluminum carbonate				
gold(II)	Au+2	nitride	N-3	Au3N2	gold(II) nitride				
cesium	Cs+1	phosphate	PO4-3	Cs3PO4	cesium phosphate				
ammonium	NH4+	sulfide	S-2	(NH4)2S	ammonium sulfide				
zinc(II)	Zn+2	nitrite	NO2-	Zn(NO2)2	zinc(II) nitrite				
chromium(I)	Cr+1	nitride	N-3	Cr3N	chromium(I) nitride				
mercury(II)	Hg+2	oxide	0-2	HgO	mercury(II) oxide				
sodium	Na+	sulfite	SO3-2	Na2SO3	sodium sulfite				
magnesium	Mg+2	oxide	0-2	MgO	magnesium oxide				

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Table of Contents Monatomic Ions Answer Key

- 1. Understanding the eBook Monatomic Ions Answer Key
 - The Rise of Digital Reading Monatomic Ions Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Monatomic Ions Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - $\circ \ \ Popular \ eBook \ Platforms$
 - Features to Look for in an Monatomic Ions Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Monatomic Ions Answer Key
 - Personalized Recommendations
 - Monatomic Ions Answer Key User Reviews and Ratings
 - Monatomic Ions Answer Key and Bestseller Lists

- 5. Accessing Monatomic Ions Answer Key Free and Paid eBooks
 - Monatomic Ions Answer Key Public Domain eBooks
 - Monatomic Ions Answer Key eBook Subscription Services
 - Monatomic Ions Answer Key Budget-Friendly Options
- 6. Navigating Monatomic Ions Answer Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Monatomic Ions Answer Key Compatibility with Devices
 - Monatomic Ions Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Monatomic Ions Answer Key
 - Highlighting and Note-Taking Monatomic Ions Answer Key
 - Interactive Elements Monatomic Ions Answer Key
- 8. Staying Engaged with Monatomic Ions Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Monatomic Ions Answer Key
- 9. Balancing eBooks and Physical Books Monatomic Ions Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Monatomic Ions Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Monatomic Ions Answer Key
 - Setting Reading Goals Monatomic Ions Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Monatomic Ions Answer Key
 - Fact-Checking eBook Content of Monatomic Ions Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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