

Lab Report 2: Characterization of Synthesis of Cadmium

Name: Elia Kufner

Matriculation No.: 21

Course: Chemie I

Reaction:

Effect of persulfate on the stability of cadmium

Time (h)	Conc. of Cd^{2+} (mg/L)	Conc. of $\text{S}_2\text{O}_8^{2-}$ (mg/L)
0	10.0	10.0
1	10.0	10.0
2	10.0	10.0
3	10.0	10.0
4	10.0	10.0
5	10.0	10.0
6	10.0	10.0
7	10.0	10.0
8	10.0	10.0
9	10.0	10.0
10	10.0	10.0
11	10.0	10.0
12	10.0	10.0
13	10.0	10.0
14	10.0	10.0
15	10.0	10.0
16	10.0	10.0
17	10.0	10.0
18	10.0	10.0
19	10.0	10.0
20	10.0	10.0
21	10.0	10.0
22	10.0	10.0
23	10.0	10.0
24	10.0	10.0
25	10.0	10.0
26	10.0	10.0
27	10.0	10.0
28	10.0	10.0
29	10.0	10.0
30	10.0	10.0
31	10.0	10.0
32	10.0	10.0
33	10.0	10.0
34	10.0	10.0
35	10.0	10.0
36	10.0	10.0
37	10.0	10.0
38	10.0	10.0
39	10.0	10.0
40	10.0	10.0
41	10.0	10.0
42	10.0	10.0
43	10.0	10.0
44	10.0	10.0
45	10.0	10.0
46	10.0	10.0
47	10.0	10.0
48	10.0	10.0
49	10.0	10.0
50	10.0	10.0
51	10.0	10.0
52	10.0	10.0
53	10.0	10.0
54	10.0	10.0
55	10.0	10.0
56	10.0	10.0
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66	10.0	10.0
67	10.0	10.0
68	10.0	10.0
69	10.0	10.0
70	10.0	10.0
71	10.0	10.0
72	10.0	10.0
73	10.0	10.0
74	10.0	10.0
75	10.0	10.0
76	10.0	10.0
77	10.0	10.0
78	10.0	10.0
79	10.0	10.0
80	10.0	10.0
81	10.0	10.0
82	10.0	10.0
83	10.0	10.0
84	10.0	10.0
85	10.0	10.0
86	10.0	10.0
87	10.0	10.0
88	10.0	10.0
89	10.0	10.0
90	10.0	10.0
91	10.0	10.0
92	10.0	10.0
93	10.0	10.0
94	10.0	10.0
95	10.0	10.0
96	10.0	10.0
97	10.0	10.0
98	10.0	10.0
99	10.0	10.0
100	10.0	10.0

Reaction: Effect of persulfate on the stability of cadmium

1. Cadmium ion solution

Stock solution: 1.0 mg/L Cd^{2+} in 100 mL of 0.1 M HNO_3 (pH 1.0)

2. Cadmium ion solution (preparation)

Stock solution: 1.0 mg/L Cd^{2+} in 100 mL of 0.1 M HNO_3 (pH 1.0)

Reaction: Effect of persulfate

Effect of $\text{S}_2\text{O}_8^{2-}$ on the stability of Cd^{2+} in 0.1 M HNO_3 (pH 1.0)

Initial conc. of Cd^{2+} = 1.0 mg/L. Initial conc. of $\text{S}_2\text{O}_8^{2-}$ = 1.0 mg/L. Initial conc. of HNO_3 = 0.1 M.

Reaction: Effect of persulfate on the stability of cadmium

Stock solution: 1.0 mg/L Cd^{2+} in 100 mL of 0.1 M HNO_3 (pH 1.0). Initial conc. of $\text{S}_2\text{O}_8^{2-}$ = 1.0 mg/L. Initial conc. of HNO_3 = 0.1 M.

Reaction: Effect of persulfate on the stability of cadmium

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E Durkheim



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