

Results and Calculations:

→ Copper Salt:

• Calculations A:

	Trial 1	Trial 2	Trial 3
Mass of sample (g)	0.703 g	0.702 g	0.7011 g
Mass of filter paper (g)	1.092 g	1.100 g	1.0701 g
Mass of filter paper + Copper precipitate (g)	1.403 g	1.309 g	1.3040 g
Mass of Copper precipitate (g)	0.311 g	0.209 g	0.2335 g
Percent of Copper in original sample (%)	44.24 %	34.045 %	33.36 %

• Calculations B:

	Calculations 
Average percent Cu in original sample (%)	$\frac{x_1 + x_2 + x_3}{3} = \frac{44.24 + 34.045 + 33.36}{3} = 37.20\%$
Standard deviation	$S = \sqrt{\frac{\sum (x_i - \bar{x})^2}{n - 1}} = \sqrt{\frac{7.84^2 + 3.16^2 + 3.84^2}{3 - 1}} = 4.895$
Relative Standard deviation	$\frac{S}{\text{mean}} = \frac{4.895}{37.20} = 0.1318$
Reported percent copper	mean $\pm$ S = 37.20 $\pm$ 4.895

• Observations:

Observations 
The reason of adding water on the precipitate after filtration? to get rid of excess HCl
The colour of the copper precipitate? blue

Gravimetric Determination Of Copper Lab Report

Thomas Griffiths

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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