

QUESTION 8

In order to investigate the relationship between the molecular mass of various fluoroalkanes and their boiling point, Jerome collects the data shown in the table below.

Haloalkane	Molecular mass (u)	Boiling point ($^{\circ}\text{C}$)
CH_3F	34	-78
$\text{C}_2\text{H}_5\text{F}$	48	-38
$\text{C}_3\text{H}_7\text{F}$	76	33
$\text{C}_4\text{H}_9\text{F}$	90	63

- 8.1 Identify the predominant intermolecular force in CH_3F . (1)
- 8.2 Is fluorobutane a GAS or a LIQUID at room temperature (25°C)? (1)
- 8.3 Plot a graph of boiling point versus molecular mass for the given data on the axes provided on your ANSWER SHEET. Draw a straight line of best fit. (6)
- 8.4 Jerome wants to use the data he collected to determine the relationship between molecular mass and boiling point. USE THE GRAPH that you have drawn in 8.3 to determine the boiling point of fluorobutane.

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S Ashworth

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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