

# Part B-1

Answer all questions in this part.

Directions (31–50): For each statement or question, record on your separate answer sheet the number of the word or expression that, of those given, best completes the statement or answers the question. Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry.

31 What is the number of electrons in an  $\text{Al}^{3+}$  ion?

- (1) 10 (3) 3  
(2) 13 (4) 16

Al +13p

32 The valence electron of which atom in the ground state has the greatest amount of energy?

- (1) cesium (3) rubidium  
(2) lithium (4) sodium

33 The numbers of protons and neutrons in each of four different atoms are shown in the table below.

Protons and Neutrons in Four Different Atoms

Atom	Number of Protons	Number of Neutrons
A	8	8
D	9	9
E	9	10
G	10	10

Which two atoms represent isotopes of the same element?

- (1) A and D (3) E and D  
(2) A and G (4) E and G

34 Which elements have the most similar chemical properties?

- (1) boron and carbon  
(2) oxygen and sulfur  
(3) aluminum and bromine  
(4) argon and silicon

35 Which element reacts with oxygen to form ionic bonds?

- (1) calcium (3) chlorine  
(2) hydrogen (4) nitrogen

36 The table below gives the atomic mass and the abundance of the two naturally occurring isotopes of chlorine.

Naturally Occurring Isotopes of Chlorine

Isotopes	Atomic Mass of the Isotope (u)	Natural Abundance (%)
$^{35}\text{Cl}$	34.97	75.76
$^{37}\text{Cl}$	36.97	24.24

Which numerical setup can be used to calculate the atomic mass of the element chlorine?

- (1)  $(34.97 \text{ u})(75.76) + (36.97 \text{ u})(24.24)$   
(2)  $(34.97 \text{ u})(0.2424) + (36.97 \text{ u})(0.7576)$   
(3)  $(34.97 \text{ u})(0.7576) + (36.97 \text{ u})(0.2424)$   
(4)  $(34.97 \text{ u})(24.24) + (36.97 \text{ u})(75.76)$

37 Which general trends in first ionization energy and electronegativity values are demonstrated by Group 15 elements as they are considered in order from top to bottom?

- (1) The first ionization energy decreases and the electronegativity decreases.  
(2) The first ionization energy increases and the electronegativity increases.  
(3) The first ionization energy decreases and the electronegativity increases.  
(4) The first ionization energy increases and the electronegativity decreases.

38 An aluminum sample has a mass of 80.01 g and a density of 2.70 g/cm<sup>3</sup>. According to the data, to what number of significant figures should the calculated volume of the aluminum sample be expressed?

- (1) 1 (3) 3  
(2) 2 (4) 4

$$D = \frac{M}{V}$$

$$V = \frac{M}{D}$$

$$\frac{80.01 \text{ (4)}}{2.70 \text{ (3)}}$$

# Geometry Regents Answers For June 2014

**SJ Ball**



## Geometry Regents Answers For June 2014:

*Regents Exams and Answers: Geometry, Sixth Edition* Barron's Educational Series, Andre, Ph.D. Castagna, 2025-01-07  
Barron's Regents Exams and Answers Geometry provides essential review for students taking the Geometry Regents including actual exams administered for the course thorough answer explanations and comprehensive review of all topics This edition features Three actual administered Regents exams so students can get familiar with the test plus one new sample test for the most recent exam changes for June 2025 Fully revised and up to date review and practice grouped by topic to help refresh skills learned in class Thorough explanations for all answers Score analysis charts to help identify strengths and weaknesses Study tips and test taking strategies All geometry topics are covered such as basic angle and segment relationships parallel lines polygons triangle relationships constructions transformations triangle congruence and writing proofs similarity and right triangle geometry parallelograms circles and arcs coordinate geometry and volume modeling 3 D shapes in practical applications Publisher's Note Products purchased from 3rd party sellers are not guaranteed by the publisher for quality authenticity or access to any online entities included with the product **Bulletin**, 1900

**Documents of the Senate of the State of New York** New York (State). Legislature. Senate, 1901 **High School Department Bulletins** University of the State of New York, 1900 Contains proceedings of various teachers associations academic examination papers etc **Slow Burn** R. Jisung Park, 2025-07-29 How the subtle but significant consequences of a hotter planet have already begun from lower test scores to higher crime rates and how we might tackle them today It's hard not to feel anxious about the problem of climate change especially if we think of it as an impending planetary catastrophe In *Slow Burn* R. Jisung Park encourages us to view climate change through a different lens one that focuses less on the possibility of mass climate extinction in a theoretical future and more on the everyday implications of climate change here and now Drawing on a wealth of new data and cutting edge economics Park shows how climate change headlines often miss some of the most important costs When wildfires blaze what happens to people downwind of the smoke When natural disasters destroy buildings and bridges what happens to educational outcomes Park explains how climate change operates as the silent accumulation of a thousand tiny conflagrations imperceptibly elevated health risks spread across billions of people pennies off the dollar of productivity fewer opportunities for upward mobility By investigating how the physical phenomenon of climate change interacts with social and economic institutions Park illustrates how climate change already affects everyone and may act as an amplifier of inequality Wealthier households and corporations may adapt quickly but without targeted interventions less advantaged communities may not Viewing climate change as a slow and unequal burn comes with an important silver lining It puts dollars and cents behind the case for aggressive emissions cuts and helps identify concrete steps that can be taken to better manage its adverse effects We can begin to overcome our climate anxiety Park shows us when we begin to tackle these problems locally **Mathematical Problem Solving** ALAN H. SCHOENFELD, 2014-06-28

This book is addressed to people with research interests in the nature of mathematical thinking at any level to people with an interest in higher order thinking skills in any domain and to all mathematics teachers The focal point of the book is a framework for the analysis of complex problem solving behavior That framework is presented in Part One which consists of Chapters 1 through 5 It describes four qualitatively different aspects of complex intellectual activity cognitive resources the body of facts and procedures at one's disposal heuristics rules of thumb for making progress in difficult situations control having to do with the efficiency with which individuals utilize the knowledge at their disposal and belief systems one's perspectives regarding the nature of a discipline and how one goes about working in it Part Two of the book consisting of Chapters 6 through 10 presents a series of empirical studies that flesh out the analytical framework These studies document the ways that competent problem solvers make the most of the knowledge at their disposal They include observations of students indicating some typical roadblocks to success Data taken from students before and after a series of intensive problem solving courses document the kinds of learning that can result from carefully designed instruction Finally observations made in typical high school classrooms serve to indicate some of the sources of students' often counterproductive mathematical behavior

Annual Report University of the State of New York. High School Department, 1901

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**Forthcoming Books** Rose Arny, 2003

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**Geometry** Helen R. Pearson, James E. Lightner, 1984

**Regents Geometry Power Pack Revised Edition** Barron's Educational Series, Andre, Ph.D. Castagna, 2021-01-05

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Kaplan,1982-11-01      **Geometry Regents Course Workbook** Donny Brusca,2020-02-21 Course Workbook for the New York State Geometry Regents Exam Each section contains key terms and concepts model problems calculator instructions and practice problems Answer key and supplemental text of Regents exam questions available separately at CourseWorkbooks.com

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