

# 8 • Electron Configurations & Periodicity

## WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
Element: Ar # of e <sup>-</sup> s: <u>18</u>	Element: Mg # of e <sup>-</sup> s: <u>12</u>	Element: N # of e <sup>-</sup> s: <u>7</u>	Element: Li # of e <sup>-</sup> s: <u>3</u>	Element: P # of e <sup>-</sup> s: <u>15</u>	Element: Cl # of e <sup>-</sup> s: <u>17</u>

Write the electron configurations of each of these in **long form** and **short form**:

- Ar  $1s^2 2s^2 2p^6 3s^2 3p^4$   
Ar  $[Ne] 3s^2 3p^4$
- Mg  $1s^2 2s^2 2p^6 3s^2$   
Mg  $[Ne] 3s^2$
- N  $1s^2 2s^2 2p^3$   
N  $[He] 2s^2 2p^3$
- Li  $1s^2 2s^1$   
Li  $[He] 2s^1$
- P  $1s^2 2s^2 2p^4 3s^2 3p^3$   
P  $[Ne] 3s^2 3p^3$
- Cl  $1s^2 2s^2 2p^6 3s^2 3p^5$   
Cl  $[Ne] 3s^2 3p^5$



- Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.



# Electron Configuration Answers

**Jiben Roy**



## Electron Configuration Answers:

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*Saraswati Chemistry Class 10* RP Manchanda, A text book on Chemistry

**A Complete Preparation for the MCAT.: Knowledge & comprehension of science** James L. Flowers, 1988

## Reviewing **Electron Configuration Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Electron Configuration Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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