8 • Electron Configurations & Periodicity

WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2	3.	4.	5.	6.
	30000 30°C				S S S S S S S S S S S S S S S S S S S
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1	¹	1	189	'⊗	'₩
Element: Ar # of e 's: 18	Element: Mg # of e 's: 12	Element: N # of e ⁻ 's: ⁷	Element: Li # of e-'s: _3	Element: P # of e 's: 15	Element: Cl # of e 's: 17

Write the electron configurations of each of these in long form and short form:

- Ar 152 252 206 352 306
 - AT TNET 352 3pt
- Mg 152 252 206 352
 - Mg [Ne] 352
- N 152 252 2p3 N [He] 252 2p3
- Li 152 25'
- P 152 252 Zpc 352 3p3
 - P [Ne] 35+ 3p3
- CI 152 252 2p6 352 3p5
- CI [Ne] 352 305



Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.

Fe 1s2 252 2p 352 3p6 3d6 452

Fe [Ar] 3d 452



Electron Configuration Practice Answer Key Table

Liying Dong

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Electron Configuration Practice Answer Key Table**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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