Key

Electron Configuration Evaluation

	ne space below, write the unabbreviated electron configurations of wring elements:	of the
1)	sodium /5 2 25 2 p 35 1	
2)	iron 152 252 2pt 352 3p6 4523	bd www. www. Wan
3)	bromine 152 252 2p6 3523p6 45230	110 4 20 30 11/1/16/1-1
4)	barium 152252p6 3523p6 4523d104p6	5524/1057 652
5)	neptunium 15 25 2p 435 3p 45 3 30 4p 55 4 40 10 51	106524F145410 BP
	he space below, write the abbreviated electron configurations of ments:	the following
6)	cobalt LAr 45 3d	
7)	silver	
8)	tellurium EKr 55° 401° 5p4	
9)	radium IRn 752	
10)) lawrencium [Rn] 752 5F 14	
Det	termine what elements are denoted by the following electron cor-	nfigurations:
11)) 1s ² 2s ² 2p ⁶ 3s ² 3p ⁴	
12)	1 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ¹ Rb	
13)) [Kr] 5s ² 4d ¹⁰ 5p ³ <u>A.5</u>	
14)	(Xe) 6s ² 4f ¹⁴ 5d ⁶ 0.5	
15)	5) [Rn] 7s ² 6d ¹ 5f ¹¹	
The second	etermine which of the following electron configurations are not	did:
Del	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s (d ¹)p ⁵	
2	2	
\approx		
(18	8) (Ra)7s ² 5f ⁸	
19		
20	o) [Xe]Xe	

Electron Configuration And Periodicity Review Answer Key

Tao Wei

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