

# Test Unit 3 SCH3U #10

Name: \_\_\_\_\_

$\frac{\text{K}}{15}$   $\frac{\text{T}}{20}$   $\frac{\text{C}}{5}$   $\frac{\text{A}}{7}$  Assessment of Learning (Product)

Multiple Choice: Circle the most correct answer. (10K)

1. What is the molar mass of  $\text{AuCl}_3$ ?

- A) 96 g                      B) 130 g                      C) 232.5 g                      D) 303.6 g

2. Avogadro's number of representative particles is equal to one \_\_\_\_\_.

- A) kilogram    B) gram                      C) Kelvin                      D) mole

3. How many moles of tungsten atoms are in  $4.8 \times 10^{22}$  atoms of tungsten?

- A)  $8.0 \times 10^2$  moles    B)  $8.0 \times 10^1$  moles    C)  $1.3 \times 10^{-1}$  moles    D)  $1.3 \times 10^{-2}$  moles

4. Determine the mass in grams of 0.0489 mol cobalt.

- A) 0.00812 g    B) 0.294 g                      C) 2.88 g                      D) 0.000830 g

5. What is the mass in grams of  $1.02 \times 10^{24}$  atoms manganese (Mn)?

- A)  $0.112 \times 10^1$  g                      B)  $0.169 \times 10^6$  g                      C)  $9.30 \times 10^{-1}$  g                      D)  $9.30 \times 10^6$  g

6. Which of the following is/are true?

- i. Samples of water coming from different sources will still have the same percentage composition
- ii. The percentage composition of a compound can ONLY be calculated with knowledge of the compound's formula
- iii. Percent composition is the relative volume of each element in the compound
- iv. Two compounds can have identical percentage compositions

- A) i, ii                      B) i, ii, iii                      C) ii, iii                      D) i, iv                      E) iv only

7. Which of the following is a candidate for an empirical formula?

- A)  $\text{C}_{24}\text{H}_{40}\text{O}_{22}$     B)  $\text{N}_2\text{O}_4$                       C)  $\text{CH}_3\text{OH}$                       D)  $\text{C}_6\text{H}_6$                       E)  $\text{Na}_2\text{S}_2\text{O}_8$

8. The mass percent of nitrogen, in nitric acid,  $\text{HNO}_3$ , is

- A) 1.6%                      B) 11.1%                      C) 15.6%                      D) 22.2%

9. Hydrogen peroxide is known to contain 5.9% hydrogen and 94.1% oxygen, by mass. If its molar mass is 34.0 g/mol, its molecular formula is

- A)  $\text{HO}$                       B)  $\text{H}_2\text{O}_2$                       C)  $\text{HO}_2$                       D)  $\text{HO}_3$

# Grade 11 Stoichiometry Practice Test

**D Keegan**



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web mar 24 2022 j analyse la manière dont le film dérange et reconstruit les idées de mort dans le cinéma de la crise du sida par la reconnaissance de l'individu séropositif en tant que quelqu'un qui ne sera pas déssexualisé par le virus malgré le fait qu'il est mourant

la petite mort 2009 imdb - Jan 10 2023

web oct 10 2009 horror thriller three friends simon nina and dodo travel to mallorca but their flight is delayed in frankfurt accidentally they get mugged and wind up at a restaurant called maison de la petite mort a notorious venue where members of the elite class secretly meet in the basement for blood read all director marcel walz writers

**la petite mort wikipedia** - Aug 17 2023

web la petite mort french pronunciation la p ə tit mɔʁ the little death is an expression that means the brief loss or weakening of consciousness and in modern usage refers specifically to the sensation of post orgasm as likened to death 1

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web oct 22 2017 les épisodes 1 à 9 de la s2 sont sur france tv sur france tv slash la petite mort at medium custom2 at campaign 800 at custom1 description at

*la petite morte wikipedia* - Feb 28 2022

web la petite morte is a 2003 canadian documentary directed by emmanuelle schick garcia about the pornography business in france centering on the interviews of raffaella anderson john b root and others it won three film festival awards for best documentary and one nomination for best documentary citation needed the title is a reference to la

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web dans son sens le plus courant aujourd'hui l'expression petite mort souvent précédé de une désigne un événement dans la vie d'une personne qui marque la fin d'une période qui la rapproche de la fin cette personne est toujours en