

Elementary Principles of Chemical Processes Solutions Manual 3rd Update Edition Edition Richard M. Felder

CHAPTER TWO

2.2 (a) $\frac{1.25 \text{ kg}}{1 \text{ min}} \left(\frac{1 \text{ hr}}{60 \text{ min}} \right) \left(\frac{1000 \text{ g}}{1 \text{ kg}} \right) = \frac{20.8 \text{ kg}}{1 \text{ hr}} \left(\frac{1000 \text{ g}}{1 \text{ kg}} \right) = 20,800 \text{ g/hr}$

(b) $\frac{0.5 \text{ kg} \cdot \text{hr}^{-1}}{1 \text{ hr}} \left(\frac{1000 \text{ g}}{1 \text{ kg}} \right) \left(\frac{1 \text{ min}}{60 \text{ sec}} \right) = \frac{8.33 \text{ g}}{1 \text{ min}} \left(\frac{1 \text{ hr}}{60 \text{ min}} \right) = \frac{0.139 \text{ g}}{1 \text{ sec}}$

(c) $\frac{0.04 \text{ cm}^3}{1 \text{ kg}} \left(\frac{1 \text{ g}}{1000 \text{ kg}} \right) \left(\frac{1 \text{ hr}}{60 \text{ min}} \right) \left(\frac{1000 \text{ g}}{1 \text{ kg}} \right) = \frac{6.67 \times 10^{-7} \text{ cm}^3}{1 \text{ min}} \left(\frac{1 \text{ hr}}{60 \text{ min}} \right) = \frac{1.11 \times 10^{-8} \text{ cm}^3}{1 \text{ sec}}$

2.3 (a) $\frac{100 \text{ mm}}{1 \text{ s}} \left(\frac{1 \text{ m}}{1000 \text{ mm}} \right) \left(\frac{1 \text{ hr}}{3600 \text{ s}} \right) = \frac{2.78 \text{ m}}{1 \text{ hr}}$

(b) $\frac{100 \text{ mm}}{1 \text{ hr}} \left(\frac{1 \text{ m}}{1000 \text{ mm}} \right) \left(\frac{1 \text{ s}}{3600 \text{ hr}} \right) = \frac{2.78 \text{ m}}{1 \text{ hr}}$

(c) $\frac{1.27 \times 10^{-7} \text{ kg}}{1 \text{ s}} \left(\frac{1 \text{ min}}{60 \text{ s}} \right) \left(\frac{1000 \text{ g}}{1 \text{ kg}} \right) \left(\frac{1 \text{ hr}}{3600 \text{ min}} \right) = \frac{3.53 \times 10^{-7} \text{ g}}{1 \text{ hr}}$

2.4 Assume that a golf ball (mass = 0.045 kg) is launched from a 2 m × 2 m × 2 m volume. If we assume atmospheric density (1.2 kg/m³) is 1.2 kg/m³.

$m_{\text{air}} = 0.045 \text{ kg} \times 1.2 \text{ kg/m}^3 \left(\frac{0.002 \text{ m}}{1 \text{ m}} \right)^3 \left(\frac{1 \text{ m}^3}{1000 \text{ L}} \right) = 0.04 \times 10^{-6} \text{ kg} = 0.04 \text{ } \mu\text{g}$

The estimate could vary by an order of magnitude or more, depending on the assumptions made.

2.5 (a) Right: $\frac{1 \text{ kg}}{1 \text{ s}} \left(\frac{1 \text{ min}}{60 \text{ s}} \right) \left(\frac{1000 \text{ g}}{1 \text{ kg}} \right) \left(\frac{1 \text{ hr}}{3600 \text{ min}} \right) = \frac{1 \text{ g}}{1 \text{ hr}}$

(b) Wrong: does not result in the units = 1000000000

$\frac{1000000000}{1 \text{ s}} \left(\frac{1 \text{ hr}}{3600 \text{ min}} \right) \left(\frac{1000 \text{ g}}{1 \text{ kg}} \right) \left(\frac{1 \text{ min}}{60 \text{ s}} \right) = 1 \text{ g}$

2.6 (a) $\frac{1000 \text{ mm}}{1 \text{ s}} \left(\frac{1 \text{ m}}{1000 \text{ mm}} \right) \left(\frac{1 \text{ hr}}{3600 \text{ s}} \right) = \frac{0.278 \text{ m}}{1 \text{ hr}}$

(b) $\frac{1000 \text{ mm}}{1 \text{ hr}} \left(\frac{1 \text{ m}}{1000 \text{ mm}} \right) \left(\frac{1 \text{ s}}{3600 \text{ hr}} \right) = \frac{0.278 \text{ m}}{1 \text{ hr}}$

Calculate the total cost for steel in dollars.

Total Cost (steel) = $14.4 \text{ kg/ton} \times \frac{0.1 \text{ kg}}{1 \text{ g}} \left(\frac{1 \text{ g}}{1000 \text{ kg}} \right) \times 1000000 = 14.4 \text{ kg} \times 1000000 = 14,400,000 \text{ kg}$

Total Cost (steel) = $14,400,000 \text{ kg} \times \frac{0.25 \text{ kg}}{1 \text{ g}} \left(\frac{1 \text{ g}}{1000 \text{ kg}} \right) \times 1000000 = 3,600,000,000 \text{ kg}$

Repeat the two steps for steel in kg

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Katrin Zwirglmaier



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