| 4. | Explain how the location of the electron bonds in Model 2 might account for the incredible hardness of a diamond. | 7 |
|----|--|---|
| 5. | Explain how the location of the "electron bonds" in Model 3 might account for the softness and flexibility of copper metal. | ı |
| 6. | Both sugar (C ₆ H ₁₂ O ₆) and salt (NaCl) dissolve in water but sugar does not conduct electricity. Use the models to explain how these two substances might be different once they dissolve in water. | |
| 7. | Use the models to explain the following evidences | |
| | a. Molecular covalent substances are often gasses. | |
| | b. Ionic substances are brittle | |

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