

1. There are four types of orbitals:
- s :** shaped like a sphere
An E level can contain only 1 s orbital, making up the "s sublevel".
 - p :** shaped like dumbbells
An E level can contain 3 p orbitals, making up the "p sublevel".
 - d :** shaped like double dumbbells
An E level can contain 5 d orbitals, making up the "d sublevel".
 - f :** too complex to draw or describe
An E level can contain 7 f orbitals, making up the "f sublevel".
2. Each orbital can hold a maximum of 2 electrons. Since both electrons have a negative charge, they repel. What keeps them from flying apart?
Each electron spins on its axis. One spins clockwise and the other spins counter-clockwise. When charged particles spin, they act like tiny magnets. Since the two electrons spin in opposite directions, one acts like the north pole of a magnet and the other acts like the south pole. This makes the electrons attract.
3. Since each orbital can hold 2 electrons:
- The "s sublevel" can hold 2 electrons.
 - The "p" sublevel can hold 6 electrons.
 - The "d" sublevel can hold 10 electrons.
 - The "f" sublevel can hold 14 electrons.

We use this notation to describe an electron:



How are electrons distributed within a sublevel?

According to Hund's Rule, each orbital within a sublevel is half-filled before any is filled.

Electron Distributions Review Answer Key

O García



Electron Distributions Review Answer Key:

Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Electron Distributions Review Answer Key**

In some sort of inundated with monitors and the cacophony of immediate conversation, the profound power and psychological resonance of verbal artistry frequently disappear into obscurity, eclipsed by the constant barrage of sound and distractions. However, located within the musical pages of **Electron Distributions Review Answer Key**, a captivating work of fictional brilliance that impulses with natural thoughts, lies an unique trip waiting to be embarked upon. Written by a virtuoso wordsmith, this exciting opus guides visitors on a psychological odyssey, lightly exposing the latent possible and profound influence embedded within the complicated internet of language. Within the heart-wrenching expanse of this evocative examination, we will embark upon an introspective exploration of the book is central themes, dissect their charming publishing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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