The Significance of Electrons in the Transportation of Energy in Living Systems

In histogical systems, Condution and Reduction Reactions are united in the production and transfer of energy. Usually, as Chadadon Reaction removes electrons from one compound and transfer name to another compound. This addition of electrons to another compound is called Reduction Reaction. These two reactions occur aincultanceously, and it is referred to as Redox Reaction (Laboreous, 2022).

The transfer of electrons between compounds or molecules a susportant. This is because most of the energy stored in atoms and deploy in faciling cell functions is normally "in the form of high-energy electrons". When this high energy is transferred in the form of electrons, it enables the cell to transport and white the energy in an incremental number rather than in a magic, destroutive faction (Libertents, 2002).

Migror carriers of electron are derived from Wramin B group, and they are the derivatives of nucleon des, and they are easily reducible and out-fixing compounds (Libertern, 2022).

Typical example includes Nicotinamide Adequae Dissultentide, which is bloomated as NAD. The NAD has two rings, and one of them is the succtinamide ring, which is the active part of the molecule that accepts and denates two electrons. Nicotinamide Adequae Dissultentide serves as an electron acceptor during the process of glycolysis, and the syule of once and of cellular respiration and denates name to confusive phosphorylation (Becker, n.4.).

In conclusion, distring are votal in the transportation of energy between molecules

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Shasha Hu

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