

1. The use of a color indicator up to the point of color to detect if the change has occurred. The color of the solution changes if the change has occurred.
2. The use of a color indicator to detect if the change has occurred. The color of the solution changes if the change has occurred.
3. The use of a color indicator to detect if the change has occurred. The color of the solution changes if the change has occurred.
4. The use of a color indicator to detect if the change has occurred. The color of the solution changes if the change has occurred.
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9. The use of a color indicator to detect if the change has occurred. The color of the solution changes if the change has occurred.
10. The use of a color indicator to detect if the change has occurred. The color of the solution changes if the change has occurred.

CONCLUSION

Write about your learning about the properties of the elements listed.

When did you see in the experiment? If something worked out, you should have noticed that the color change occurred with an orange-yellow color. When the color change occurred, it was a change in the color, a change in the color of the solution. The color change should have been purple, and if you did the step using the color change, it should have been purple.

When did you see in the experiment? If something worked out, you should have noticed that the color change occurred with an orange-yellow color. When the color change occurred, it was a change in the color, a change in the color of the solution. The color change should have been purple, and if you did the step using the color change, it should have been purple.

think about this



Now, at the end of this section of the book, we have three chemical reactions that show the change in the color of the solution. The color of the solution changes if the change has occurred.

ON YOUR OWN

1. Write about the following elements: sodium, potassium, calcium, magnesium, and aluminum. Write about the color change in the solution.
2. Explain the difference between the color change in the solution and the color change in the solution.

THE PERIODIC TABLE OF THE ELEMENTS

The periodic table of the elements (Figure 8.1) is a chemical's most important tool. It shows the elements in a way that is easy to understand. The elements are arranged in a way that shows their properties. The elements are arranged in a way that shows their properties. The elements are arranged in a way that shows their properties.

FIGURE 8.1
The Periodic Table of the Elements

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Hussin A.Rothana



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