



Name: Sharanya Saxena

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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Since there are 8 markers in a set, Nate would have to give one marker to Clara.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give up two markers, one of which can go to each Fred and Florence.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (▶) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

2. Click **Pause** (⏸). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

- A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Sodium

Which is a nonmetal?

Chlorine

- B. Try moving an electron from the metal to the nonmetal. What happens?

Sodium lets go of its one electron.

Exploration Guide Covalent Bonds Answer Key

Ian Pickup



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Teacher's Manual and Resource Guide for Exploring the Sciences Herbert Drapkin,1964 *Student Guide for Cycles of Life* Gerarld L. Kellogg,2006 Prentice Hall Science Explorer Michael J. Padilla,2002 **Resources for Teaching Middle School Science** Smithsonian Institution,National Academy of Engineering,National Science Resources Center of the National Academy of Sciences,Institute of Medicine,1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents

Computational Approaches for Chemistry Under Extreme Conditions Nir Goldman,2019-02-18 This book presents recently developed computational approaches for the study of reactive materials under extreme physical and thermodynamic conditions It delves into cutting edge developments in simulation methods for reactive materials including quantum

calculations spanning nanometer length scales and picosecond timescales to reactive force fields coarse grained approaches and machine learning methods spanning microns and nanoseconds and beyond These methods are discussed in the context of a broad range of fields including prebiotic chemistry in impacting comets studies of planetary interiors high pressure synthesis of new compounds and detonations of energetic materials The book presents a pedagogical approach for these state of the art approaches compiled into a single source for the first time Ultimately the volume aims to make valuable research tools accessible to experimentalists and theoreticians alike for any number of scientific efforts spanning many different types of compounds and reactive conditions **Lehninger Principles of Biochemistry** David L. Nelson,Albert L. Lehninger,Michael M. Cox,2008-02 Authors Dave Nelson and Mike Cox combine the best of the laboratory and best of the classroom introducing exciting new developments while communicating basic principles of biochemistry **Chemical Interactions** ,2005 **Holt Chemistry** Ralph Thomas Myers,2004 **The Software Encyclopedia 2000** Bowker Editorial Staff,2000-05 **Modules** McDougal Littell Incorporated,2005 **Molecular Biology and Biotechnology** Robert Allen Meyers,1995-06-29 This is one volume library of information on molecular biology molecular medicine and the theory and techniques for understanding modifying manipulating expressing and synthesizing biological molecules conformations and aggregates The purpose is to assist the expanding number of scientists entering molecular biology research and biotechnology applications from diverse backgrounds including biology and medicine as well as physics chemistry mathematics and engineering **Library of Congress Catalog: Motion Pictures and Filmstrips** Library of Congress,1963 A cumulative list of works represented by Library of Congress printed cards **Anti-SARS-CoV-2 Activity of Flavonoids** Jen-Tsung Chen,2024-07-16 In the past years COVID 19 caused millions of deaths and severely damaged not only public health but also the global economy New variants continue to threaten human health Thus natural compounds for retarding infection are in high demand This book summarizes the emerging research with flavonoids whose application holds great promise of antiviral anti inflammation antioxidative stress and immunomodulatory Topics include the role of flavonoids in preventing an inflammatory storm and how flavonoids interact with critical protein targets that are involved in the infection stages of COVID 19 Key Features Highlights anti SARS CoV 2 drug discovery on natural products and medicinal plants Includes all major subclasses of flavonoids that promisingly combat COVID 19 Proposes molecular mechanisms of flavonoids against protein targets of SARS CoV 2 Contributions from an international team of leading researchers Provides recommendations with respect to the future research **Jacaranda Nature of Biology 2 VCE Units 3 and 4, LearnON and Print** Judith Kinnear,Marjory Martin,Lucy Cassar,Elise Meehan,Ritu Tyagi,2021-10-29 Jacaranda Nature of Biology Victoria s most trusted VCE Biology online and print resource The Jacaranda Nature of Biology series has been rewritten for the VCE Biology Study Design 2022 2026 and offers a complete and balanced learning experience that prepares students for success in their assessments by building deep understanding in both Key Knowledge and Key Science Skills Prepare students

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