

UNIT
Study Guide
Element, compounds and mixtures

QUESTIONS	ANSWERS
1. soluble	A substance that dissolves
2. condensation	The taking away of heat energy from a gas to turn into a liquid Ex. Rain
3. sublimation	Energy is being added to change a solid to a gas (skipping liquid)
4. evaporation	When a liquid changes into a gas, the process taking place at the surface of the liquid
5. physical change	-does not change the way atoms are linked up - no new substances are formed - dissolves a solute into a solvent - boiling water - melting butter -change of state may change. Ex. Water-ice to liquid
6. chemical change	- atoms change the way they link up - formation of a new substance - formation of a precipitate - producing a gas by adding 2 solutions - burning an object
7. atom	- the smallest pure part of an element - basic building block of matter
8. molecule	The smallest part of a compound that keeps the properties of that compound
9. alloy	A mixture of metals Ex. Stainless steel, jewelry
10. solute Give an example	- the substance that is being dissolved - ex. sugar
11. solvent Give an example	- the substance dissolving - ex. water

Elements Compounds And Mixtures Unit Study Guide

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Elements Compounds And Mixtures Unit Study Guide:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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