

# Electrons/Periodic Table Review Packet

Name \_\_\_\_\_

Period \_\_\_\_\_

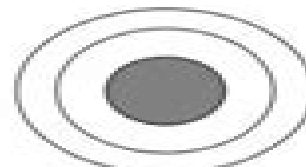
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## Atomic Basics

Name \_\_\_\_\_

### Part A: Atomic Structure

1. Draw five protons in the nucleus of the atom. Label them with their charge.
2. Draw six neutrons in the nucleus of the atom.
3. Draw two electrons in the first energy level and label them with their charge.
4. Draw three electrons in the second energy level and label them with their charge.
5. What element is represented by the diagram? \_\_\_\_\_



### Part B: Atomic Calculations

6. Label the information provided in the periodic table.

|          |         |
|----------|---------|
| 8        | ← _____ |
| <b>O</b> | ← _____ |
| Oxygen   | ← _____ |
| 15.999   | ← _____ |

7. What does the atomic number represent?

\_\_\_\_\_ or \_\_\_\_\_

8. What does the atomic mass represent?

\_\_\_\_\_ + \_\_\_\_\_

9. How would you figure the number of protons or electrons in an atom?

10. How would you figure the number of neutrons in an atom?

11. Use your knowledge of atomic calculations to complete the chart.

| Element   | Atomic Number | Atomic Mass | Protons | Neutrons | Electrons |
|-----------|---------------|-------------|---------|----------|-----------|
| <b>Li</b> | 3             | 7           |         |          |           |
| <b>P</b>  | 15            | 31          |         |          |           |
| <b>Cl</b> |               | 35          | 17      |          |           |
| <b>Ni</b> | 28            |             |         | 34       |           |
| <b>K</b>  |               | 39          |         |          | 19        |
| <b>Ag</b> | 47            |             |         | 61       |           |
| <b>H</b>  |               | 1           | 1       |          |           |
| <b>Si</b> |               |             |         | 14       | 14        |
| <b>W</b>  |               |             | 74      | 108      |           |
| <b>Ne</b> |               |             |         | 10       | 10        |

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## Part C: Electron Configuration

# Electrons And The Periodic Table Review Answers

**JG Myers**



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## **Table of Contents Electrons And The Periodic Table Review Answers**

1. Understanding the eBook Electrons And The Periodic Table Review Answers
  - The Rise of Digital Reading Electrons And The Periodic Table Review Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Electrons And The Periodic Table Review Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Electrons And The Periodic Table Review Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Electrons And The Periodic Table Review Answers
  - Personalized Recommendations
  - Electrons And The Periodic Table Review Answers User Reviews and Ratings
  - Electrons And The Periodic Table Review Answers and Bestseller Lists

5. Accessing Electrons And The Periodic Table Review Answers Free and Paid eBooks
  - Electrons And The Periodic Table Review Answers Public Domain eBooks
  - Electrons And The Periodic Table Review Answers eBook Subscription Services
  - Electrons And The Periodic Table Review Answers Budget-Friendly Options
6. Navigating Electrons And The Periodic Table Review Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Electrons And The Periodic Table Review Answers Compatibility with Devices
  - Electrons And The Periodic Table Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Electrons And The Periodic Table Review Answers
  - Highlighting and Note-Taking Electrons And The Periodic Table Review Answers
  - Interactive Elements Electrons And The Periodic Table Review Answers
8. Staying Engaged with Electrons And The Periodic Table Review Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Electrons And The Periodic Table Review Answers
9. Balancing eBooks and Physical Books Electrons And The Periodic Table Review Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Electrons And The Periodic Table Review Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Electrons And The Periodic Table Review Answers
  - Setting Reading Goals Electrons And The Periodic Table Review Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Electrons And The Periodic Table Review Answers
  - Fact-Checking eBook Content of Electrons And The Periodic Table Review Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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