

Periodic Table Groups and Periods

A group is a column on the periodic table. Elements have the same number of valence electrons.

A period is a row on the periodic table. Elements have the same number of electron shells.

There are 7 periods.

There are 18 groups.

The periodic table is color-coded by groups: Group 1 (blue), Group 2 (yellow), Groups 3-10 (orange), Group 11 (light blue), Group 12 (light green), Groups 13-18 (light blue), and Groups 19-20 (light purple). The lanthanide and actinide series are shown below the main table.

1	2																	18																					
1	H																	2	He																				
3	Li	4	Be																	10	Ne																		
5	Na	6	Mg																	16	Ar																		
7	K	8	Ca	9	Sc	11	Ti	12	V	13	Cr	14	Mn	15	Fe	16	Co	17	Ni	18	Cu	19	Zn	20	Ga	21	Ge	22	As	23	Se	24	Br	25	Kr				
11	Rb	12	Sr	13	Y	14	Zr	15	Nb	16	Mo	17	Tc	18	Ru	19	Rh	20	Pd	21	Ag	22	Cd	23	In	24	Sn	25	Sb	26	Te	27	I	28	Xe				
17	Cs	18	Ba	19	La	20	Hf	21	Ta	22	W	23	Re	24	Os	25	Ir	26	Pt	27	Au	28	Hg	29	Tl	30	Pb	31	Bi	32	Po	33	At	34	Rn				
23	Fr	24	Ra	25	Ac	26	Rf	27	Db	28	Sg	29	Bh	30	Hs	31	Mt	32	Ds	33	Rg	34	Cn	35	Nh	36	Fl	37	Mc	38	Lv	39	Ts	40	Og				
35	La	36	Ce	37	Pr	38	Nd	39	Pm	40	Sm	41	Eu	42	Gd	43	Tb	44	Dy	45	Ho	46	Er	47	Tm	48	Yb	49	Lu										
41	Ac	42	Th	43	Pa	44	U	45	Np	46	Pu	47	Am	48	Cm	49	Bk	50	Cf	51	Es	52	Fm	53	Md	54	No	55	Lr										

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