8 • Electron Configurations & Periodicity

WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
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2000000	2 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	2000	2000 gr	20000	2000000 C
' ©	' ⊠	¹₩	180	180	'₩
# of e 's: 10	# of e 's: 12	Element: N # of e 's: 7	Element: Li # of e-'s: _3	Element: P # of e 's: 15	Element: Cl # of e 's: 17

Write the electron configurations of each of these in long form and short form:

- Ar 152 252 206 352 306 AT [NE] 352 3p6
- Mg 152 252 204 352 Mg TNE 7 352
- 3. N 152 252 2p3 N [He] 252 2p3
- Li [152 25'
- P 152 252 Zp 352 3p3 P [Ne] 35+ 3p3
- CI 152 252 2p6 352 3p5
- CI [Ne] 352 305



Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.

Fe 1s2 2s2 2p 3s2 3p6 3d6 452

Fe [Ar] 3d 4 452

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Heather Hattori, Richard H. Langley

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