

## Notes:

- i. All questions are compulsory.
- ii. Answers to the two sections are to be written in the same answer book.
- iii. Figures to the right hand side indicate full marks.
- iv. Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- v. Every new question must be started on a new page.
- vi. Use of logarithmic table is allowed.

## CHEMISTRY: SECTION – I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- i. In body centred cubic structure the space occupied is about  
(A) 68 % (B) 53 %  
(C) 38 % (D) 32 %
- ii. For a gaseous reaction the unit of rate of reaction is  
(A)  $\text{L atm s}^{-1}$  (B)  $\text{atm mol}^{-1} \text{s}^{-1}$   
(C)  $\text{atm s}^{-1}$  (D)  $\text{mol s}$
- iii. Which of the following compounds contains S – O as well as S – S bonds?  
(A) Sulphuric acid (B) Thiosulphuric acid  
(C) Sulphurous acid (D) Thiosulphurous acid
- iv. Which of the following solutions shows maximum depression in freezing point?  
(A) 0.5 M  $\text{Li}_2\text{SO}_4$  (B) 1 M NaCl  
(C) 0.5 M  $\text{Al}_2(\text{SO}_4)_3$  (D) 0.5 M  $\text{BaCl}_2$
- v. For a chemical reaction  $\Delta S = -0.035 \text{ kJ/K}$  and  $\Delta H = -20 \text{ kJ}$ .  
(A) 5.14 K (B) 57.14 K  
(C) 571.4 K (D) 5714.0 K
- vi. The standard e.m.f of the following cell is 0.463 V  
 $\text{Cu} | \text{Cu}_{(0.50)}^{++} || \text{Ag}_{(0.50)}^+ | \text{Ag}$ . If  $E_{\text{Ag}}^{\circ} = 0.800 \text{ V}$ ,  
What is the standard potential of Cu electrode?  
(A) 1.137 V (B) 0.337 V  
(C) 0.463 V (D) -0.463 V
- vii.  $\text{Fe}_2\text{O}_3$  is reduced to spongy iron near the top of blast furnace by  
(A)  $\text{H}_2$  (B) CaO  
(C)  $\text{SiO}_2$  (D) CO

# Hsc Questions 2013

**RJ Shavelson**



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## **Table of Contents Hsc Questions 2013**

1. Understanding the eBook Hsc Questions 2013
  - The Rise of Digital Reading Hsc Questions 2013
  - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Questions 2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hsc Questions 2013
  - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Questions 2013
  - Personalized Recommendations
  - Hsc Questions 2013 User Reviews and Ratings
  - Hsc Questions 2013 and Bestseller Lists
5. Accessing Hsc Questions 2013 Free and Paid eBooks

- Hsc Questions 2013 Public Domain eBooks
  - Hsc Questions 2013 eBook Subscription Services
  - Hsc Questions 2013 Budget-Friendly Options
6. Navigating Hsc Questions 2013 eBook Formats
    - ePub, PDF, MOBI, and More
    - Hsc Questions 2013 Compatibility with Devices
    - Hsc Questions 2013 Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Hsc Questions 2013
    - Highlighting and Note-Taking Hsc Questions 2013
    - Interactive Elements Hsc Questions 2013
  8. Staying Engaged with Hsc Questions 2013
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Hsc Questions 2013
  9. Balancing eBooks and Physical Books Hsc Questions 2013
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Hsc Questions 2013
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Hsc Questions 2013
    - Setting Reading Goals Hsc Questions 2013
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Hsc Questions 2013
    - Fact-Checking eBook Content of Hsc Questions 2013
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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