

Honors Chemistry
Practice Test #4

Name _____
Date _____

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- Which one of the following compounds is not a salt?
 - Ag_2SO_4
 - Ca(OH)_2
 - KCl
 - $\text{Ca(NO}_3)_2$
 - $\text{NaC}_2\text{H}_3\text{O}_2$
- Which of the following is the BEST definition of a salt?
 - A substance that produces ions in solution.
 - A substance that is completely ionized in solution.
 - A compound composed of metals and nonmetals.
 - A compound that contains a cation other than H^+ and an anion other than OH^- .
 - A substance that does not produce ions in solution.
- Which one of the following compounds is a salt?
 - HClO_2
 - $\text{HC}_2\text{H}_3\text{O}_2$
 - HCN
 - $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$
 - KOH
- Which one of the following is a strong acid?
 - HF
 - HNO_2
 - $\text{HC}_2\text{H}_3\text{O}_2$
 - H_2SO_3
 - H_2CO_3
- Which statement regarding nitric acid is false?
 - It only slightly ionizes in solution.
 - Its solutions conduct electricity.
 - It is soluble in water.
 - It is a strong electrolyte.
 - It produces H^+ and NO_3^- in solution.
- Which one of the following is a weak acid?
 - HBr
 - HNO_2
 - HClO_4
 - HCl
 - HClO_2
- Which one of the following is a strong and soluble base?
 - Ni(OH)_2
 - Mg(OH)_2
 - BaOH
 - Ca(OH)_2
 - Fe(OH)_3

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