Holt Chemistry – Guided Notes, Chapter 1: The Science of Chemistry Section 1: What is Chemistry? Objectives:

- Describe ways in which chemistry is a part of your daily life.
- Describe the characteristics of three common states of matter.
- Describe physical and chemical changes, and give examples of each.
- Identify the reactants and products in a chemical reaction.
- List four observations that suggest a chemical change has occurred.

A _____ is any substance that has a definite composition. A _____ is the process by which one or more substances change to produce one or more new substances. The _____ of ____ are the physical forms of matter which are solid, liquid, gas, and plasma.

- have a fixed volume and shape that result from the way their particles are arranged.
- have a fixed volume but not a fixed shape.
 have neither fixed volume nor shape.

Water molecule, H₂O Water molecule, H₂O Water molecule, H₂O

Physical changes are changes in which the

of a substance doesn't change.

Changes of state are physical changes.

Chemical changes occur when the ______ of substances change and new substances form.

Evidence of a chemical change:



without accepta, secol, or vinegast, and soldinam hydrogen carbonate, or baking soda, are mircol, the solution bubbles as carbon dionale forms.



b. Where solitations of solitans solitate and endinners intrate are mixed, codmissis solitate, a solid proceputate, forms.



 When aluminum nearly with iron(EH) inside in the clay post, emergy is released in heart and light.



d When phenolphthalein is added to announce dissolved in water, a color change from coloritos to pink occurs.

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SA Dillow

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