

Date: No.

4. A Set of Standard Solutions to Establish a Standardization Curve
Molar concentration of $\text{Fe}(\text{NO}_3)_3$, 0.002 M, Molar concentration of NaOCl , 0.002 M

Standard Solution	React	Fe^{3+}	I^-	I_2	I^-	I_2
A1. Volume of NaOCl (mL)	0 mL	1 mL	2 mL	3 mL	4 mL	5 mL
A2. Molar of OCl^- (mol)						
A3. $[\text{OCl}^-]$ (0.01 M)						
A4. $[\text{Fe}(\text{NO}_3)_3]$ (mol/L)						
A5. Percent transmittance, %T (for molar readings only)						
A6. Absorbance, A	0.007	0.134	0.193	0.445	0.571	0.656

*Calculation for Standard Solution 1.

A7. Plot data of A versus $[\text{Fe}(\text{NO}_3)_3]$, determine λ_{max} is approx

Experiment 34 Equilibrium Constant Report Sheet

Answers

Albert A Gayle



Experiment 34 Equilibrium Constant Report Sheet Answers:

Reviewing **Experiment 34 Equilibrium Constant Report Sheet Answers**: Unlocking the Spellbinding Force of Linguistics

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