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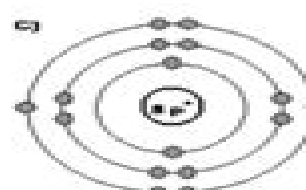
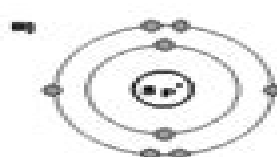
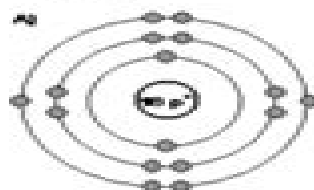
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PRETEST ON CHAPTER 1
"ATOMIC STRUCTURE & THE PERIODIC TABLE OF ELEMENTS"

Part 1 – MULTIPLE CHOICES

Answer all the questions on the multiple choice sheet provided at the end

- 1) Oxygen is a gas important for life and it represents about 21% of the Earth's atmosphere. Which of the illustrations below best represents a Rutherford-Bohr diagram of an oxygen atom?



- 2) Listed below are the characteristics of an element from the periodic table:

- It is a non metal
- Its outermost level has 7 electrons
- It is used to purify and disinfect water

To which group in the periodic table does this element belong?

- A) Alkali metals B) Alkaline earth metals C) Halogens D) Inert gases

- 3) Listed below are the characteristics of an element from the periodic table:

- It is a metal
- It has only one valence electron
- It is stored in oil due to its high chemical reactivity

To which group in the periodic table does this element belong?

- A) Alkali metals B) Alkaline earth metals C) Halogens D) Inert gases

- 4) Listed below are the characteristics of an element from the periodic table:

- Its outermost level has 8 electrons
- It does not react with any other elements

Elements And The Periodic Table Pretest

Kathleen Armour



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