**企材 東 原原** 

### **Extension Questions**

- 24. The hydrogen spectral lines in Model 2 are only the wavelengths of light that are in the visible range and therefore "seen" by the naked eye. However, many other wavelengths can be detected with special equipment.
  - a. Propose a hydrogen electron transition that involves light with a wavelength in the ultraviolet (UV) range (10–400 nm).

wavelengths within the UV range would have more energy than violet wavelengths. This transition would be from n=6 to n=2 or may be n=6 to n=1.

 Propose a hydrogen electron transition that involves light with a wavelength in the infrared (IR) range (1000–106 nm).

wavelengths within the IR range would have less energy than red wavelengths. This transition would be from n=3 to n=2 or maybe u=5 to n=4.

 Below are diagrams for the bright line spectra of four elements and the spectrum of a mixture of unknown gases.

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# **Electron Energy And Light Extension Questions Pogil Answers**

**Christian Drosten** 

**Electron Energy And Light Extension Questions Pogil Answers:** 

The Enigmatic Realm of **Electron Energy And Light Extension Questions Pogil Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Electron Energy And Light Extension Questions Pogil Answers** a literary masterpiece penned with a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of people who partake in its reading experience.

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