8 • Electron Configurations & Periodicity

WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
, <u>M</u> <u>M A A</u>	, 2000	.2000	,2000 1,2000	8000	S S S S S S S S S S S S S S S S S S S
2000 X	300000 C	2000	20000 P	2000 O	2000000 200000000000000000000000000000
1	'₩	¹₩	'Ø	¹⊗	¹₩
Element: Ar # of e 's: 18	Element: Mg # of e 's: 12	Element: N # of e 's: 7	Element: Li # of e-'s: _3	Element: P # of e 's: 15	Element: Cl # of e 's: 17

Write the electron configurations of each of these in long form and short form:

- Ar 152 252 206 352 306
 - AT [NE] 352 3p6
- Mg 152 252 2pt 352
 - Mg TNE 7 352
- 3. N 152 252 2p3 N [He] 252 2p3
- Li [152 25'
- 5. P 152 252 Zp4 352 3p3
 - P [Ne] 35+ 3p3
- CI 152 252 2p6 352 3p5
 - CI [Ne] 352 305



Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.

Fe 1st 2st 2pt 352 3pt 3d 452

Fe [Ar] 3d 4 452



Electron Configuration Review Answer Sheet

AW Rasmussen

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