

CHM 101

Objectives:

- Explain the nature of light and how it reveals the electron structure of the atom.
- Write the electron structure of elements
- Relate electron structure to the periodic table

1. Light is a form of what type of energy?
 - Give 2 examples of other forms of this type of energy.
2. What is the meaning of wavelength and frequency of light?
3. White light produces a continuous spectrum. Elements that are excited produce a line spectrum. What is a line spectrum?
4. Different metals emit different colored light in flame tests. This light originates from inside the metal atoms. Explain what was happening inside these atoms that causes the various colors of light. Be specific.
5. What is an orbital?
6. List the orbitals that exist in the following energy levels. Indicate the number of these orbitals.
 $n = 1$, $n = 2$, $n = 3$, $n = 4$
7. Indicate the electron configuration of these elements using the shorthand notation and the orbital diagrams.

For example - helium: $1s^2$, $1s$

Elements: U, C, Mg, Fe, Kr

8. In this periodic chart, indicate the outer orbitals being filled with electrons in the rectangles.

9. Why do elements in the same group have similar properties?

Electron Configuration Worksheet Packet

J Elliott

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

Electron Configuration Worksheet Packet:

Electron Configuration Worksheet Packet Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Electron Configuration Worksheet Packet**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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