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ELECTROCHEMISTRY

MULTIPLE CHOICE QUESTIONS

									•			
1.	Reduction always takes place:										KLHR 05,	1.20
	(m)	At anode					(b)	At cat	thode			
	(c)	At both electro	des				(d)	does	not occu	r at electr	rodes	
2.	The electrolyte KOH is used in cell:										(GRW	06)
	(a) Lead accumulator						(b)	MI - C	id cell			
	(c) Alkaline battery						(d)	Silver oxide battery				
3.	In the reaction 2Fe + 3Cl ₂ +2FeCl ₃ :								(Blww	P 10, GRV	W 09, LHR	06)
	(m)	Fig. is reduced					(b)	Fig. is a	beautibisco			700
	(c)	Cl ₂ is oxidized					(d)	None	of these	happens		
4.	Oxidation state of hydrogen in CaH ₂ is:									Out of the Contract of	(GRW 0	773
	(m)	+1	(b)	-1			(c)	+2	(4)	zeno		
5.	Cu metal can be purified in electrolytic cell by making the impure Cu as: (GRW 07)											
	(m)	Anode (b)	Cathoo	te(c)	Anode	aind Cal	thode	(4)	Deper	nds upon	nature of	solution
6.	Loss of electrons is called:											171
	(a)	Oxidation	(b)	Reduc	dikory	(c)	Hydrat	tikom	(d)	Dehydr	ation	
7.	Fuel cells convert chemical energy into:											0.70
	(40)	Heat energy	(b)	Light -	energy	(c)	Electric	cal ener	ngnyr ((d))	Mechanic	cal energy	
8.	Electrolysis is used for:										06)	
	(a)	Electroplating	(b)	Pfamut	facture of	sodium	n emetal	(c)	Manua	facture of	Al (d) /	All of these
9.	Nelson's cell and Down's cell are example of: (ESD 09, GRW 08)											0(8)
	(a)	Electrochemica	if cett	(b)	Galwan	ile celli	(c)	Electr	olytic ce	(d) B	None of t	these
10.	The oxidation number of Cr in K ₂ Cr ₂ O ₂ is:								CORNA	09, 10, L	HR 10,12,	13)
	(a)	+14	(b)	+12		(c)	+6		(d)	+13		
11.	The e	The electrolyte used in fuel cell is:								LHIR O'P, C	HGH, GRW	10)
	East.	American Black	436.5	Ballan Street	m. AdmiCE	Keth.	AND COLUMN		J. 43.	Manker		

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