

Reactions in Aqueous Solutions: Metathesis Reactions and Net Ionic Equations

9 Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

1. Write molecular, complete ionic, and net ionic equations for the reactions that occur, if any, when solutions of the following substances are mixed:
 - (a) nitric acid and potassium carbonate
 - (b) sodium bromide and lead nitrate
 - (c) acetic acid and calcium hydroxide
 - (d) calcium nitrate and sodium sulfate
 - (e) ammonium chloride and lithium hydroxide
2. Which of the following are not water-soluble: $\text{Ba}(\text{NO}_3)_2$, FeCl_3 , CuCO_3 , CuSO_4 , ZnS , and ZnSO_4 ?
3. Write equations for the decomposition of H_2CO_3 and H_2SO_3 .
4. At what temperature (from your graph) do KNO_3 and NaCl have the same molar solubility?

Metathesis Reaction Lab Answers

Neil D. Jespersen, Alison Hyslop



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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order **Fundamentals of Chemistry in the Laboratory** Ralph A. Burns, 1999 Energy Research Abstracts , 1986 **ACT Science Prep Course** Jeff Kolby, 2016-01-01 Comprehensive Prep for ACT Science Every year students pay 1 000 and more to test prep companies to prepare for the science section of the ACT Now you can get the same preparation in a book Although the ACT science section is difficult it is very learnable ACT Science Prep Course presents a thorough analysis of ACT science and introduces numerous analytic techniques that will help you immensely not only on the ACT but in college as well The ACT cannot be beaten But it can be mastered through hard work analytical thought and by training yourself to think like a test writer Many of the exercises in this book are designed to prompt you to think like an ACT test writer Features Comprehensive Review Fifteen chapters provide complete review of basics of ACT science Practice Includes 75 examples 280 problems and 240 test questions Full length Tests Six full length tests will thoroughly prepare you for the test Performance If your target is a top score this is the book Chemistry James E. Brady, Fred Senese, 2004-02-04 Publisher Description **BIS Exam PDF-Technical Assistant (Lab) Chemical eBook PDF** Chandresh Agrawal, nandini books, 2024-06-12 SGN The eBook BIS Technical Assistant Lab Chemical Covers Chemistry Subject Objective Questions From Various Exams With Answers **Chemistry, Study Guide** James E. Brady, John R. Holum, 1993-02-11 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples *Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Synthetic and Mechanistic Studies of a Vanadium-catalyzed Enantioselective Oxidation of**

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