

## AP Chemistry Review

### Chapter 6 and 7, Atomic Structure and Periodicity

- Be familiar with the electromagnetic spectrum
  1. What types of light are shorter in length than microwave? Know the whole order of the spectrum.
  2. What type of light helps scientists to learn about electron structure?
  3. What creates the emission lines in a spectra for an element? What would cause only certain wavelengths of light to be absorbed by an atom?
- Convert between frequency and wavelength
  4. Calculate the frequency of light emitted by exciting Sr atoms if the wavelength is 461 nm.
  5. Calculate the wavelength of light emitted by exciting Na atoms if the frequency is  $3.45 \times 10^{14} \text{ s}^{-1}$
- Calculate the energy of a wave or an electron
  6. Calculate the energy of 1 mole of photons emitted by a Li atom if the wavelength is 670nm.
- Define and understand:
  7. Ground state, excited state, Pauli exclusion principle, Heisenberg Uncertainty Principle, Hund's rule.
- Know the maximum number of electrons in and shape of each type of orbital.
  8. 3s                      5p                      4f                      6d
- Following the Aufbau principle, determine electron configurations, both longhand and condensed, for elements and ions
- Match each explanation below with the correct letter
  - (A)  $1s^2 2s^2 2p^5 3s^2 3p^5$
  - (B)  $1s^2 2s^2 2p^6 3s^2 3p^6$
  - (C)  $1s^2 2s^2 2p^6 2d^{10} 3s^2 3p^6$
  - (D)  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5$
  - (E)  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 4s^2$
- 9. An impossible electron configuration
- 10. The ground-state configuration for the atoms of a transition element
- 11. The ground-state configuration of a negative ion of a halogen
- 12. The ground-state configuration of a common ion of an alkaline earth element
- 13. List 3 species that are isoelectronic
- 14. Which of the following represents the ground state electron configuration for the  $\text{Mn}^{3+}$  ion? (Atomic number Mn = 25)
  - (a)  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^4$
  - (b)  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 4s^2$
  - (c)  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 4s^2$
  - (d)  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 4s^2$
  - (e)  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 4s^1$
- Draw Dot diagrams for elements
  15. Boron                      Calcium                      Krypton

# Merrill Chemistry Study Guide Chapter 6 Answers

**Emilie Sanchez**



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*Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998      *Merrill Earth Science* Ralph M. Feather, Susan Leach Snyder, Dale T. Hesser, 1995      **Vocational-technical Learning Materials** Bruce Reinhart, 1974  
*Choice* Richard K. Gardner, Phyllis Grumm, 1976      **Teaching Chemistry - A Studybook** Ingo Eilks, Avi Hofstein, 2013-04-20 This book focuses on developing and updating prospective and practicing chemistry teachers pedagogical content knowledge The 11 chapters of the book discuss the most essential theories from general and science education and in the second part of each of the chapters apply the theory to examples from the chemistry classroom Key sentences tasks for self assessment and suggestions for further reading are also included The book is focused on many different issues a teacher of chemistry is concerned with The chapters provide contemporary discussions of the chemistry curriculum objectives and assessment motivation learning difficulties linguistic issues practical work student active pedagogies ICT informal learning continuous professional development and teaching chemistry in developing environments This book with contributions from many of the world s top experts in chemistry education is a major publication offering something that has not previously been available Within this single volume chemistry teachers teacher educators and prospective teachers will find information and advice relating to key issues in teaching such as the curriculum assessment and so forth but contextualised in terms of the specifics of teaching and learning of chemistry and drawing upon the extensive research in the field Moreover the book is written in a scholarly style with extensive citations to the literature thus providing an excellent starting point for teachers and research students undertaking scholarly studies in chemistry education whilst at the same time offering insight and practical advice to support the planning of effective chemistry teaching This book should be considered essential reading for those preparing for chemistry teaching and will be an important addition to the libraries of all concerned with chemical education Dr Keith S Taber University of Cambridge Editor Chemistry Education Research and Practice The highly regarded collection of authors in this book fills a critical void by providing an essential resource for teachers of chemistry to enhance pedagogical content knowledge for teaching modern chemistry Through clever orchestration of examples and theory and with carefully framed guiding questions the book equips teachers to act on the relevance of essential chemistry knowledge to navigate such challenges as context motivation to learn thinking activity language assessment and maintaining professional expertise If you are a secondary or post secondary teacher of chemistry this book will quickly become a favorite well thumbed resource Professor Hannah Sevan University of Massachusetts Boston  
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population of learners It features down to earth language clear explanations and specific examples of abstract concepts The author examines the relationship between learning and other topics of importance to educators development motivation instruction and self regulation A detailed glossary of more than 300 technical terms facilitates understanding and a bibliography with more than 1 100 references encourages self study For future and in service teachers with a minimal psychology background *Working Mother* ,2002-10 The magazine that helps career moms balance their personal and professional lives **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office,1969

*Introduction to Epidemiology* Ray M. Merrill,Thomas C. Timmreck,2006 An Introduction to Epidemiology Fourth Edition is intended for introductory courses in health related programs at both the advanced undergraduate and graduate levels It is also a valuable reference for epidemiologists working in the field industrial hygienists infectious disease nurses and staff epidemiologists **Final Causality in Nature and Human Affairs** Richard F. Hassing,2018-03-02 Teleology the inquiry into the goals or goods at which nature history God and human beings aim is among the most fundamental yet controversial themes in the history of philosophy Are there ends in nonhuman nature Does human history have a goal Do humanly unintended events of great significance express some sort of purpose Do human beings have ends prior to choice The essays in this volume address the abiding questions of final causality The chapters are arranged in historical order from Aristotle through Hegel to contemporary anthropic principle cosmology *The Publishers Weekly* ,1910 Methods for Teaching David A. Jacobsen,Paul D. Eggen,Donald P. Kauchak,2009 Developed to address the need for a text that allows teaching fundamentals to be covered in a practical fashion this general methods text provides complete and concise coverage of the teaching act with a focus on planning and implementing classroom instruction and assessing student achievement in an era of standards and accountability With an emphasis on case studies and real life classroom scenarios and features including self check exercises portfolio assignments suggested field experiences and selected tools for teaching this text provides students with the resources they need to master the material presented in the text Publisher s website Polk's Medical Register and Directory of North America ,1914 Medical and Surgical Register of the United States Comprising ... Index of All the Physicians in the United States ... ,1914 **Teaching and Learning in the Elementary School** Judy Reinhartz,Don M. Beach,1997 This comprehensive realistic view of teaching and curriculum development in the contemporary elementary school uses an interactive reflective and research based approach supported with numerous practical explanations examples and illustrations Written in easy to understand language it 1 covers the historical philosophical social and psychological background of teaching and curriculum development 2 outlines the steps and strategies for planning implementing and assessing the curriculum and for teaching throughout construction of traditional or thematic teaching units and 3 offers an overview of the various subject areas taught in the elementary school concepts goals resources and tools Practicing elementary school teachers

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