

## Mark Scheme

Q1.

| Question Number | Acceptable Answer                                                                                                                                                                                          | Additional Guidance                                                                                                                                                                                                                                                                                                        | Mark |
|-----------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------|
| (a) (i)         | $\text{CaCO}_3(\text{s}) + 2\text{HCl}(\text{aq}) \rightarrow \text{CaCl}_2(\text{aq}) + \text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g})$<br><br>Balanced equation<br>(1)<br><br>State symbols<br>(1) | Accept<br>$\text{CaCO}_3(\text{s}) + 2\text{H}^+(\text{aq}) \rightarrow \text{Ca}^{2+}(\text{aq}) + \text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g})$<br><br>2nd mark dependent on first or near miss.<br><br>Reject $\text{H}_2\text{CO}_3(\text{aq})$ in equation, but allow state symbol mark if otherwise correct. | (2)  |
| Question Number | Acceptable Answer                                                                                                                                                                                          | Additional Guidance                                                                                                                                                                                                                                                                                                        | Mark |
| (a) (ii)        | Finds molar mass of calcium carbonate                                                                                                                                                                      | Example of calculation<br>$\text{Mr of calcium carbonate}$<br>$= 40.1 + 12 + (16 \times 3) = 100.1 \text{ (g mol}^{-1}\text{)}$<br>Allow<br>$= 40 + 12 + (16 \times 3) = 100 \text{ (g mol}^{-1}\text{)}$<br>Accept answer with no working                                                                                 | (1)  |

# May 2013 Chemistry Mark Schemeexcel As

**M Tight**



**May 2013 Chemistry Mark Schemeexcel As:**

## Unveiling the Power of Verbal Beauty: An Mental Sojourn through **May 2013 Chemistry Mark Scheme** excel As

In a world inundated with displays and the cacophony of instantaneous conversation, the profound energy and psychological resonance of verbal artistry often diminish in to obscurity, eclipsed by the regular assault of noise and distractions. However, nestled within the musical pages of **May 2013 Chemistry Mark Scheme** excel As, a captivating perform of fictional beauty that impulses with organic emotions, lies an unique journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, that interesting opus books visitors on an emotional odyssey, lightly revealing the latent potential and profound influence stuck within the delicate internet of language. Within the heart-wrenching expanse with this evocative examination, we can embark upon an introspective exploration of the book is key subjects, dissect its interesting publishing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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