

1

[3]

F51

161

[3]

The fundamental ideas in chemistry

electron
 nucleus
 neutron
 each for 1 mark

2.

(m)

- B D
- (ii) D
 (b) A and B only one type of atom
 C and D more than one type of atom
 accept element for atom
 ignore the word 'mixaire'
 (chemically) bonded

accept (chemically) joined or similar idea of joined

- 3. B
 carbon monoxide
 1

 CO
 accept carbon oxide
 1

 do not credit carbon dioxide
 1
 - do not credit if any superscripts or subscripts used but accept C₁O₁, accept OC do not credit if obviously lower case
 - C water
 H₂O accept hydrogen oxide
 do not accept hydrogen hydroxide
 do not credit if obviously lower case or if 2 not subscript
 do not accept HOH
 accept OHs
 - D ammonia do not accept ammonium 1
 NH3 do not credit if obviously lower case, or if 3 not subscript 1
 accept nitrogen hydride or hydrogen nitride
 do not accept hydrogen nitrate or nitrite
 allow H₃N
- 4. correct reactants (i.e. sodium + water either way round)
 - correct products (i.e. sodium hydroxide + hydrogen, either way round)
- correct use of 'react'/'reaction'/reactants'/combine (not mixed'added/join) 3
 correct use of 'produces'/products'/gives/forms/makes/creates reactants
 correctly identified

 each for 1 mark
 - products correctly identified



1

May 13 C1 Markscheme

Mer Merlyn

May 13 C1 Markscheme:

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Table of Contents May 13 C1 Markscheme

- 1. Understanding the eBook May 13 C1 Markscheme
 - The Rise of Digital Reading May 13 C1 Markscheme
 - Advantages of eBooks Over Traditional Books
- 2. Identifying May 13 C1 Markscheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - ∘ Features to Look for in an May 13 C1 Markscheme
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from May 13 C1 Markscheme
 - Personalized Recommendations

- May 13 C1 Markscheme User Reviews and Ratings
- May 13 C1 Markscheme and Bestseller Lists
- 5. Accessing May 13 C1 Markscheme Free and Paid eBooks
 - o May 13 C1 Markscheme Public Domain eBooks
 - May 13 C1 Markscheme eBook Subscription Services
 - May 13 C1 Markscheme Budget-Friendly Options
- 6. Navigating May 13 C1 Markscheme eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - May 13 C1 Markscheme Compatibility with Devices
 - May 13 C1 Markscheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of May 13 C1 Markscheme
 - Highlighting and Note-Taking May 13 C1 Markscheme
 - Interactive Elements May 13 C1 Markscheme
- 8. Staying Engaged with May 13 C1 Markscheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers May 13 C1 Markscheme
- 9. Balancing eBooks and Physical Books May 13 C1 Markscheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection May 13 C1 Markscheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine May 13 C1 Markscheme
 - Setting Reading Goals May 13 C1 Markscheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of May 13 C1 Markscheme
 - Fact-Checking eBook Content of May 13 C1 Markscheme

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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