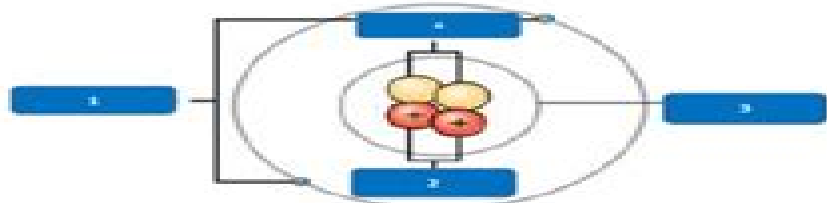


	DAILY LESSON LOG		School Batasan Hills National High School	Grade Level Learning Area	8 Chemistry																				
Teacher MARIA RIZZA E. LLORENTE		Teaching Date and Time for 50 minutes	Quarter 3rd																						
Date: March 23-24, 2023		Day: Thursday-Friday																							
DLL#68																									
I. OBJECTIVE:																									
A. Content Standard		The learners demonstrate understanding of the identity of a substance according to its atomic number																							
B. Performance Standard		The learners should be able to present how water behaves in its different states within the water cycle.																							
C. Learning Competencies		Determine the number of protons, neutrons, and electrons in a particular atom; (S8MT-IIIc-f-10)																							
Specific Objectives:		At the end of the lesson, the students should be able to: 1. Describe the atom and its sub-atomic particles; 2. Determine the number of electrons, protons, and neutrons in an atom; and 3. Explain how atoms benefit life.																							
II. CONTENT		SUBATOMIC PARTICLES Determining the number of protons, electrons and neutrons in an atom																							
III. LEARNING RESOURCES																									
A. References																									
B. Other Learning Resources		• http://chemed.chem.purdue.edu/genchem/topicreview/bp/ch6/atom_emfame.html • http://www.kentchemistry.com/links/AtomicStructure/protonsws1.htm • https://education2eb.org/lesson_number.html • https://www.nap.edu/read/4818/chapter/2 • https://www.youtube.com/watch?v=1M4jyadimRo																							
IV. PROCEDURE																									
A. Reviewing previous lesson or presenting the new lesson.		ELICIT: TRIVIA QUIZ GAME The class will play trivia quiz game about atoms and molecules																							
		https://www.canva.com/design/DAP6p7UYH0Q8kUQleQAF8isLMxgQK1CZueA/edit																							
B. Establishing a purpose for the lesson		ENGAGE: LABEL THE PICTURE: What's Inside the Atom?																							
																									
		a. Ask the class to label the picture. b. Introduce subatomic particles. c. Recall the proponent/discovery of the atomic model on the picture.																							
		Answer: 1. Electron 2. Proton 3. Nucleus 4. neutron																							
C. Presenting examples/instances of the new lesson		ENGAGE: TABLE ANALYSIS The students will study the table below and answer the questions about Properties of Subatomic Particles.																							
		<table><tr><th>Subatomic particle</th><th>Symbol</th><th>Relative charge</th><th>Location</th><th>Mass</th></tr><tr><td>Proton</td><td>p</td><td>+1</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Neutron</td><td>n</td><td>0</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Electron</td><td>e⁻</td><td>-1</td><td>Outside the nucleus</td><td>0.0005 (-1/1836)</td></tr></table>				Subatomic particle	Symbol	Relative charge	Location	Mass	Proton	p	+1	Inside the nucleus	1	Neutron	n	0	Inside the nucleus	1	Electron	e ⁻	-1	Outside the nucleus	0.0005 (-1/1836)
Subatomic particle	Symbol	Relative charge	Location	Mass																					
Proton	p	+1	Inside the nucleus	1																					
Neutron	n	0	Inside the nucleus	1																					
Electron	e ⁻	-1	Outside the nucleus	0.0005 (-1/1836)																					

Lesson Plan Chapter 22 Subatomic

Tao Wei



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Table of Contents Lesson Plan Chapter 22 Subatomic

1. Understanding the eBook Lesson Plan Chapter 22 Subatomic
 - The Rise of Digital Reading Lesson Plan Chapter 22 Subatomic
 - Advantages of eBooks Over Traditional Books
2. Identifying Lesson Plan Chapter 22 Subatomic
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lesson Plan Chapter 22 Subatomic
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lesson Plan Chapter 22 Subatomic
 - Personalized Recommendations
 - Lesson Plan Chapter 22 Subatomic User Reviews and Ratings
 - Lesson Plan Chapter 22 Subatomic and Bestseller Lists
5. Accessing Lesson Plan Chapter 22 Subatomic Free and Paid eBooks
 - Lesson Plan Chapter 22 Subatomic Public Domain eBooks
 - Lesson Plan Chapter 22 Subatomic eBook Subscription Services
 - Lesson Plan Chapter 22 Subatomic Budget-Friendly Options

6. Navigating Lesson Plan Chapter 22 Subatomic eBook Formats
 - ePub, PDF, MOBI, and More
 - Lesson Plan Chapter 22 Subatomic Compatibility with Devices
 - Lesson Plan Chapter 22 Subatomic Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lesson Plan Chapter 22 Subatomic
 - Highlighting and Note-Taking Lesson Plan Chapter 22 Subatomic
 - Interactive Elements Lesson Plan Chapter 22 Subatomic
8. Staying Engaged with Lesson Plan Chapter 22 Subatomic
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lesson Plan Chapter 22 Subatomic
9. Balancing eBooks and Physical Books Lesson Plan Chapter 22 Subatomic
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lesson Plan Chapter 22 Subatomic
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Lesson Plan Chapter 22 Subatomic
 - Setting Reading Goals Lesson Plan Chapter 22 Subatomic
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lesson Plan Chapter 22 Subatomic
 - Fact-Checking eBook Content of Lesson Plan Chapter 22 Subatomic
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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