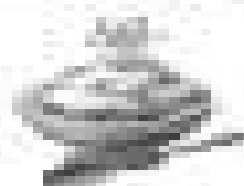


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine which elements are present in a sample of gas by observing the emission spectrum.

Part I: Making Observations in the Atomic Probing

1. Turn on the power supply to the atomic probe. Observe the emission spectrum of the gas in the tube. Record the colors of the lines in the spectrum. (The colors of the lines in the spectrum are listed in the table below.)

Observation: The emission spectrum of the gas in the tube shows a series of discrete lines of color. (Record the colors of the lines in the table below.)

2. Turn on the power supply to the atomic probe. Observe the emission spectrum of the gas in the tube. Record the colors of the lines in the spectrum. (Record the colors of the lines in the table below.)

Observation: The emission spectrum of the gas in the tube shows a series of discrete lines of color. (Record the colors of the lines in the table below.)

3. Turn on the power supply to the atomic probe. Observe the emission spectrum of the gas in the tube. Record the colors of the lines in the spectrum. (Record the colors of the lines in the table below.)

Observation: The emission spectrum of the gas in the tube shows a series of discrete lines of color. (Record the colors of the lines in the table below.)

4. Turn on the power supply to the atomic probe. Observe the emission spectrum of the gas in the tube. Record the colors of the lines in the spectrum. (Record the colors of the lines in the table below.)

Lesson 11 Atomic Pudding Answers

Sebastian Brünink



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