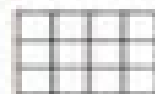


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Kenny, Kenny who is that?
Reflex in a half upon your lap?
Why it's just after I see the cat.

What is the area of this figure? ☐ is top



- A. 9
- B. 12
- C. 18
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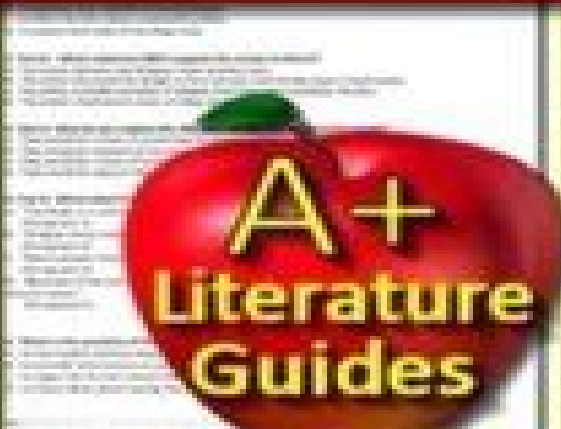


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A. The number 10 is a prime number.
B. The number 10 is a composite number.
C. The number 10 is a square number.
D. The number 10 is a cube number.

2. Which of the following is a true statement?
A. The number 10 is a prime number.
B. The number 10 is a composite number.
C. The number 10 is a square number.
D. The number 10 is a cube number.

3. Which of the following is a true statement?
A. The number 10 is a prime number.
B. The number 10 is a composite number.
C. The number 10 is a square number.
D. The number 10 is a cube number.



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Greg Faherty, Princeton Review (Firm)



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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow 2\text{Cr}^{3+} \text{aq} + 7\text{H}_2\text{O} \text{l}$

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web jan 29 2020 the ionic form of this reaction is $2\text{Na}^+ \text{aq} + 2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} \rightarrow 2\text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{s}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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 net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g} \rightarrow \text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions
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