



A-LEVEL PHYSICS

MARK SCHEME

Practice Paper 2

Maximum marks: 85

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Section A

1 (a) (i) Heating result in increased pressure because of: 2
more collisions (of molecules) with wall per second

OR more frequent collisions of molecules with walls ✓

greater momentum change (of molecules)per collision ✓

(ii) (Heating results in) faster molecules so more momentum change per collision √

(To maintain same pressure) volume increases to reduce the number of collisions per second √

(b) (i)
$$V = \left(\frac{nRT}{P}\right) = \frac{0.250 \times 8.31 \times 300}{1.05 \times 10^5} \checkmark$$
 2

 $5.9(4) \times 10^{-3} \text{ m}^3 \checkmark$

(ii)
$$T_2 = \left(\frac{T_1V_2P_2}{P_1V_1}\right) = \frac{300 \times 7.0 \times 10^6}{1.05 \times 10^5 \times 20}$$

$$\left[or \, T_2 = \left(\frac{p_2 V_2}{nR} \right) = \frac{7.0 \times 10^6 \times 5.94 \times 10^{-3}}{8.31 \times 0.25 \times 20} \right] \checkmark$$

(Total 8 marks)

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2014 Edit

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Table of Contents Hsc Physics Exam 20paper 2 Mark Scheme

- 1. Understanding the eBook Hsc Physics Exam 20paper 2 Mark Scheme
 - The Rise of Digital Reading Hsc Physics Exam 20paper 2 Mark Scheme
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Hsc Physics Exam 20paper 2 Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Physics Exam 20paper 2 Mark Scheme
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Hsc Physics Exam 20paper 2 Mark Scheme
 - Personalized Recommendations
 - Hsc Physics Exam 20paper 2 Mark Scheme User Reviews and Ratings
 - Hsc Physics Exam 20paper 2 Mark Scheme and Bestseller Lists
- 5. Accessing Hsc Physics Exam 20paper 2 Mark Scheme Free and Paid eBooks
 - ∘ Hsc Physics Exam 20paper 2 Mark Scheme Public Domain eBooks
 - Hsc Physics Exam 20paper 2 Mark Scheme eBook Subscription Services
 - Hsc Physics Exam 20paper 2 Mark Scheme Budget-Friendly Options
- 6. Navigating Hsc Physics Exam 20paper 2 Mark Scheme eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Hsc Physics Exam 20paper 2 Mark Scheme Compatibility with Devices
 - Hsc Physics Exam 20paper 2 Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Hsc Physics Exam 20paper 2 Mark Scheme
 - Highlighting and Note-Taking Hsc Physics Exam 20paper 2 Mark Scheme
 - Interactive Elements Hsc Physics Exam 20paper 2 Mark Scheme

- 8. Staying Engaged with Hsc Physics Exam 20paper 2 Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Physics Exam 20paper 2 Mark Scheme
- 9. Balancing eBooks and Physical Books Hsc Physics Exam 20paper 2 Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Physics Exam 20paper 2 Mark Scheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc Physics Exam 20paper 2 Mark Scheme
 - Setting Reading Goals Hsc Physics Exam 20paper 2 Mark Scheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc Physics Exam 20paper 2 Mark Scheme
 - o Fact-Checking eBook Content of Hsc Physics Exam 20paper 2 Mark Scheme
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - $\circ \ \ Integration \ of \ Multimedia \ Elements$
 - Interactive and Gamified eBooks

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