বিষয় কোত :	1	7	- 6
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হচনিৰ্বাচনি অভীকা

TO PROPERTY

ক্রবাদ পর

(২০২৩ সালের সিলেবাস অনুযায়ী)

সময়—২৫ মিনিট

পূৰ্ণমান—২৫

বিলেৰ দুইবা । সৰবৰাহকৃত বছনিৰ্বাচনি অভীকাৰ উত্তৰপত্তে প্ৰস্তেই ক্ৰমিক নম্বৰে বিপত্তীতে প্ৰকল্প বৰ্ণসম্পদত বৃত্তসমূহ হতে স্তিক/সৰ্বোধ্যক উত্তৰে বৃত্ততি বল পছেন্ট কলম ছাৱা সম্পূৰ্ণ তথাট কৰে। প্ৰতিটি প্ৰস্তেৱ মান ১ ।]

অপ্লশতত কোনো প্ৰকাৰ দাশ/চিফ দেৱা যাবে দা।

- চিকিৎসা বিজ্ঞানে কিঞ্জিতখেরাপিতে কোনটি ব্যবহার করা হয়?
 - (P) X-ray

সিলেট বোর্ড

- B IR
- (9) MRI
- (M) UV

উদ্দীপক থেকে ২ এবং ৩ নং প্রপ্রের উত্তর দাও :

মৌল	বহিন্ত করের ইলেবট্রন কিবাস		
Λ .	(n-1)s ¹	अकर	
В	ns²np⁴	L decision	
С	(n+1)s ² (n+1)p ⁴	मान 2	

A, B এবং C প্রচলিত থৌলের প্রতীক বর।

- ২। 🗛 ৪ ৩ 🗛 C ভৌত অবস্থার ভিন্নতার ভারণ---
 - (ক) আছনিক বছন
 - (খ) সমযোগী বন্ধল
 - হাইছেলেন বছন
 - (খ) সল্লিবেশ বন্ধন
- ৩। A2B খৌগের কেতর—
 - ে sp³ সংকলণ ফটে
 - AL বছন কোপ 104-5°
 - iii. (Lp Lp) विकर्षन विनामान

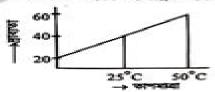
নিচের কোনটি সঠিকঃ

- (4) 14 H
- (4) II 4 III
- ৪। নিজের মৌলভলার ১ম আছনীকরণ বিভারের সঞ্জিক ক্রম কোনটিঃ
 - (*) Be>B>N>O
 - N > O > Be > B
 - (*f) O>N>B>Be (*f) B>Be>N>O

- ৫। গোটনের প্রকৃত ভর কত।
 - (*) 1-60×10⁻²⁴g
 - (4) 1-66×10⁻²⁴g
 - 9 1-673×10⁻²⁴g
 - (v) 1-675×10⁻²⁴g
- **৯। কোনটি প্রশন্ম অক্সাইড**া
 - (4) N₂O₃
- (4) NO³

12

- । S-ব্ৰক খৌলের সংখ্যা ককটিঃ
 - (**) 7 (
- **9** 14 (%) 16
- শরমাপুর ৩য় শক্তিকরে ফুর্লনরত ইলেকট্রনের কৌশিক ভরবেশ এর মান কোনটিঃ
- 3h 2×
- (四) 经
- া কোনটি উভহবৰী অপ্লাইত।
 - (1) KO₂
- (4) Na₂O₂ (4) Fe₂O₃
- উশীলক হতে ১০ এবং ১১ নং প্রপ্রের উত্তর দাও :



- ে। 200g NH₄CI এর সম্পৃত প্রকাকে 50°C থেকে 25°C তালধারার ইত্তে করলে কর প্রাথ NH₄CI কেলালিত হবে।
 - (m) 75

- (4) 50 (4) 90
 - /0 [भद्र भृक्षेत्र शुक्रेत

Hsc Chemistry Exam Mcq Question Out

Maureen Slade, Roger Slade

Hsc Chemistry Exam Mcq Question Out:

Cambridge Checkpoints HSC Chemistry 2017-19 Maureen Slade, Roger Slade, 2016-06-29 **Excel HSC Chemistry** Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every sin gle dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured. This guide comes in a brand new format that makes even better use of your study time up to date covera ge of the core topics plus 3 Option topics Industrial Chemistry Shipwr ecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to s ection the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second h and Investigations and Problem Solving %this way you will be ab le to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for word %this way you can easily match the E xcel guide to the syllabus an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work hundreds of key concept questions with answers %questions that test you recall of knowledge in each chapter HSC type quest ions for every section in each chapter with clock icons to tell you how much time you will have to answer the questions in the HSC %this way yo u can test yourself on HSC type questions under HSC type time pressure an examiner maximiser feature ticks to show the mark distribut ion and answers to all HSC type questions %all you need to answ er HSC type questions two sample HSC papers with an examiner m aximiser feature plus answers %not one but two up to date sample papers the Excel syllabus summary notes a detachable secti on at the end of the guide where every single dot point of each chapter is summarised for you% a comprehensive and compact summary of the whole course in 32 pages Excel Revise in a Month HSC Chemistry Will Marchment, 2007 **Excel HSC Chemistry** C. M. Roebuck, 2003 Excel HSC & **Preliminary Senior Science** Jennifer Hill, 2011 This comprehensive study quide covers the complete HSC Preliminary Se nior Science course and has been specifically created to maximise exam s uccess This guide has been designed to meet all study needs providing up to date information in an easy to use format The sample HSC Exam has been updated for the new format Excel HSC Preliminary Senior Science contains an introductory section including how to use the book and an explanation of the new course helpful study and exam techniques comprehensive coverage of the entir e Preliminary and HSC courses hundreds of diagrams to aid under standing icons and boxes to highlight key concepts and assessme nt skills including laboratory and field work checklists of key terms end of chapter revision questions with fully explained a nswers a trial HSC style exam with answers and explanations a glossary of key terms useful websites highlighted throu ghout

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