

# চট্টগ্রাম বোর্ড

১২।  $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$  এনিওনের অনুবর্তী ভারতক হলো—

- (ক)  $[\text{Fe}(\text{H}_2\text{O})_4(\text{OH})_2]^+$   
 (খ)  $[\text{Fe}(\text{H}_2\text{O})_5(\text{OH})]^{2+}$   
 (গ)  $[\text{Fe}(\text{H}_2\text{O})_5(\text{OH})]^{3+}$   
 (ঘ)  $[\text{Fe}(\text{H}_2\text{O})_6(\text{OH})]^{2+}$

১৩। অক্সিডেশন-রাসায়নিক কোষের emf = 7

- I.  $E^\circ_{\text{Anode (ox)}} + E^\circ_{\text{Cathode (Red)}}$   
 II.  $E^\circ_{\text{Cathode (Red)}} - E^\circ_{\text{Anode (Red)}}$   
 III.  $E^\circ_{\text{Anode (ox)}} - E^\circ_{\text{Cathode (ox)}}$

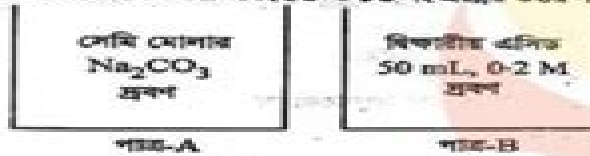
কোনটি সঠিক?

- (ক) I ও II  
 (খ) II ও III  
 (গ) I ও III  
 (ঘ) I, II ও III

১৪। 27°C তাপমাত্রায় 8 g  $\text{CH}_4$  গ্যাসের পটেনশিয়াল কত হলো?

- (ক) 935-32  
 (খ) 3741-30  
 (গ) 1870-65  
 (ঘ) 4870-30

উদ্ভিদকটি লক্ষ্য কর এবং ১৫ ও ১৬ নং প্রশ্নের উত্তর দাও।



[গ্যাস B এর প্রবণ দ্বারা A গ্যাসের প্রবণ সম্পূর্ণরূপে প্রশমিত হয়]

১৫। A গ্যাসের প্রবণের ঘনমাত্রা ppm এককে কত?

- (ক)  $5.3 \times 10^4$   
 (খ)  $1.06 \times 10^3$   
 (গ)  $9.8 \times 10^3$   
 (ঘ)  $1.96 \times 10^4$

১৬। A গ্যাসের প্রবণের আয়তন কত?

- (ক) 20 mL  
 (খ) 40 mL  
 (গ) 1000 mL  
 (ঘ) 2000 mL

১৭। বেনজিনের কার্বন-কার্বন বন্ধন দূরত্ব কত nm?

- (ক) 0.154  
 (খ) 0.139  
 (গ) 0.134  
 (ঘ) 0.120

১৮। দ্রবণে  $\text{FeSO}_4$  প্রবণ রাখলে সুই কোষের কোষ বিভব হবে—

$$[\text{E}^\circ_{\text{Zn}^{2+}(\text{aq})/\text{Zn(s)}} = -0.76 \text{ V};$$

$$[\text{E}^\circ_{\text{Fe}^{2+}(\text{aq})/\text{Fe(s)}} = -0.44 \text{ V}]$$

- (ক) -0.32 V  
 (খ) -1.20 V  
 (গ) +0.32 V  
 (ঘ) +1.20 V

১৯। কোন গ্যাসটি বয়েলের সূত্র হতে সবচেয়ে কম বিচ্যুত হবে?

- (ক)  $\text{H}_2$   
 (খ)  $\text{CO}_2$   
 (গ)  $\text{N}_2$   
 (ঘ)  $\text{O}_2$

২০। হেটেরো অ্যারোমেটিক বৌদ কোনটি?

- (ক) ইমিডিন অক্সাইড  
 (খ) বেনজিন  
 (গ) পিরিডিন  
 (ঘ) ক্রোমাইলক্সাইড

২১। কোন বৌদে 'C' এর জারণ মান শূন্য?

- (ক)  $\text{CH}_4$   
 (খ)  $\text{CO}$   
 (গ)  $\text{CH}_2\text{Cl}_2$   
 (ঘ)  $\text{C}_2\text{H}_2$

২২। একই মাত্রার বিদ্যুৎ একই সময় চালনা করলে কোন অর্ধকোষে অক্সিডেশন বেশি জমা হবে?

- (ক)  $\text{Cu}^{2+}$   
 (খ)  $\text{Fe}^{3+}$   
 (গ)  $\text{Ag}^+$   
 (ঘ)  $\text{Na}^+$

২৩। SATP তে নাইট্রাস অক্সাইড গ্যাসের ঘনত্ব  $\text{g L}^{-1}$  এককে কত?

- (ক) 1.77  
 (খ) 1.96  
 (গ) 1.85  
 (ঘ) 1.21

উদ্ভিদকটি লক্ষ্য কর এবং ২৪ ও ২৫ নং প্রশ্নের উত্তর দাও।



২৪। A বৌদটি হলো—

- (ক)  $\text{R}-\text{CH}(\text{OH})-\text{CH}_3$   
 (খ)  $\text{R}-\text{CH}_2-\text{CH}_2-\text{OH}$   
 (গ)  $\text{R}-\text{CH}_2-\text{CH}_2-\text{Br}$   
 (ঘ)  $\text{R}-\text{CH}(\text{Br})-\text{CH}_3$

২৫। উক্ত বিক্রিয়া—

- I. বারকিনিক দীর্ঘ প্রযোজ্য নয়  
 II. উৎপাদ A অসম্পূর্ণ  
 III. মুক্ত মূলক সৃষ্টির মাধ্যমে বিক্রিয়া সংঘটিত হয়

কোনটি সঠিক?

- (ক) I ও II  
 (খ) II ও III  
 (গ) I ও III  
 (ঘ) I, II ও III

# Hsc Chemistry 2nd Paper Objective Answer 23

**C. M. Roebuck**



### **Hsc Chemistry 2nd Paper Objective Answer 23:**

**CIM Bulletin** Canadian Institute of Mining, Metallurgy and Petroleum,1993      CIM Bulletin Canadian Institute of Mining and Metallurgy,Canadian Institute of Mining, Metallurgy and Petroleum,1993      *Scientific American* ,1880 Monthly magazine devoted to topics of general scientific interest      *Energy Research Abstracts* ,1989      **Paperbound Books in Print** ,1992      **Annual Meeting of the American Public Health Association and Related Organizations** American Public Health Association,1988      *Who's who in America* ,2003      **2 Unit Chemistry HSC Course Revision** New South Wales. Board of Studies,1994      *2 Unit Chemistry HSC Course Revision* ,1996      HSC Chemistry Peter Lewis,Roger Michael Slade,1986      **Excel Chemistry** C. M. Roebuck,1995      *2 Unit Chemistry Higher School Certificate Examination Paper and Answers*, 1997 Catherine Odium,1998      *2 Unit Chemistry Higher School Certificate Examination Paper and Answers*, 1996 Catherine Odium,1998      *Answers To: A Guide to HSC Chemistry* Peter Lewis,Roger Michael Slade,1984\*  
**Past HSC Chemistry Questions and Worked Answers 1986-1997 by New Syllabus Topics** Aaron Butler,1998

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Hsc Chemistry 2nd Paper Objective Answer 23**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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