Practice Problems: Double Displaceme

1. ___ Fe(s) + ___
$$H_2SO_4(aq) \rightarrow _{--} Fe_2(SO_4)_3(aq) + ___$$

2. ___ NaC₂H₃O₂(aq) + ___ H₂CO₃(aq)
$$\rightarrow$$
 ___ Na₂CO₃(aq

3. ___ (NH₄)₃PO₄(aq) + ___ Pb(NO₃)₄(aq)
$$\rightarrow$$
 ___ Pb₃(PO₄

4. ___
$$C_4H_{10}(g) + ___ O_2(g) \rightarrow ___ CO_2(g) + ___ H_2O(g)$$

5. ___
$$H_3PO_4(aq) + ___ HCl(aq) \rightarrow ___ PCl_5(s) + ___ H_2C$$

6. ___
$$Hg(OH)_2(s) + ___ H_3AsO_4(aq) \rightarrow ___ Hg_3(AsO_4)_2(s)$$

7.
$$HClO_4(aq) + P_4O_{10}(s) \rightarrow H_2PO_4(aq) +$$

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Valentine Korah

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