

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ lattice energy _____	a. ionic bond
_____ bond energy _____	b. covalent bond
_____ enthalpy of vaporization _____	c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ ionic _____	a. H—F	_____ nonpolar covalent _____	d. H—H
_____ ionic _____	b. Na—Cl	_____ polar covalent _____	e. H—C
_____ polar covalent _____	c. H—O	_____ polar covalent _____	f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction. Hydrogen bonds are generally intermolecular, while ionic and covalent bonds occur between ions or atoms respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond. Increased polarity in H₂O bonds means a stronger intermolecular attraction, making water a liquid at room temperature. Hydrogen bonding exists between water molecules, but not between hydrogen sulfide molecules.

Guided And Study Workbook Answers Covalent Bonding

RM Cervero



Guided And Study Workbook Answers Covalent Bonding:

Whispering the Strategies of Language: An Mental Quest through **Guided And Study Workbook Answers Covalent Bonding**

In a digitally-driven earth where screens reign supreme and quick interaction drowns out the subtleties of language, the profound secrets and emotional subtleties concealed within phrases frequently get unheard. However, nestled within the pages of **Guided And Study Workbook Answers Covalent Bonding** a fascinating fictional treasure sporting with organic emotions, lies a fantastic journey waiting to be undertaken. Composed by a skilled wordsmith, that charming opus invites viewers on an introspective journey, lightly unraveling the veiled truths and profound affect resonating within the very fabric of each and every word. Within the mental depths of the emotional review, we can embark upon a honest exploration of the book is primary themes, dissect its charming publishing design, and succumb to the strong resonance it evokes serious within the recesses of readers hearts.

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