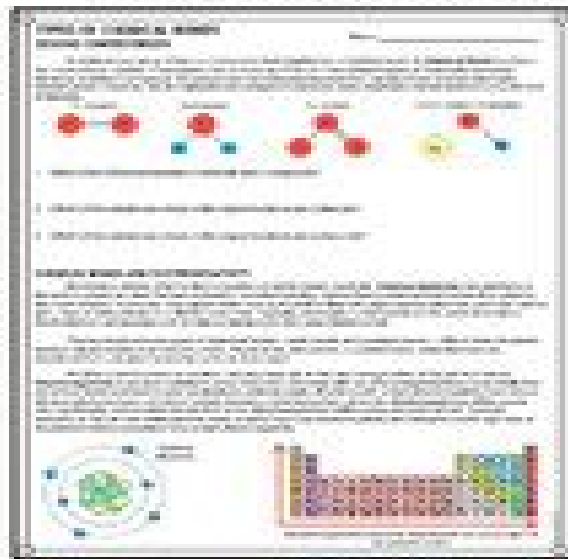


GUIDED READING WITH CHUNKED QUESTIONS THROUGHOUT & POST-READING QUESTIONS



Types of Chemical Bonds

1. Ionic bonds are formed between a metal and a non-metal. The metal atom loses an electron to become a positive ion, and the non-metal atom gains an electron to become a negative ion. The resulting ions are attracted to each other by electrostatic forces, forming an ionic bond.

2. Covalent bonds are formed between two non-metal atoms. The atoms share a pair of electrons, forming a covalent bond.

3. Metallic bonds are formed between metal atoms. The metal atoms are arranged in a regular lattice, and the outer electrons are delocalized, forming a 'sea of electrons' that holds the lattice together.

Diagram showing the formation of ionic, covalent, and metallic bonds.



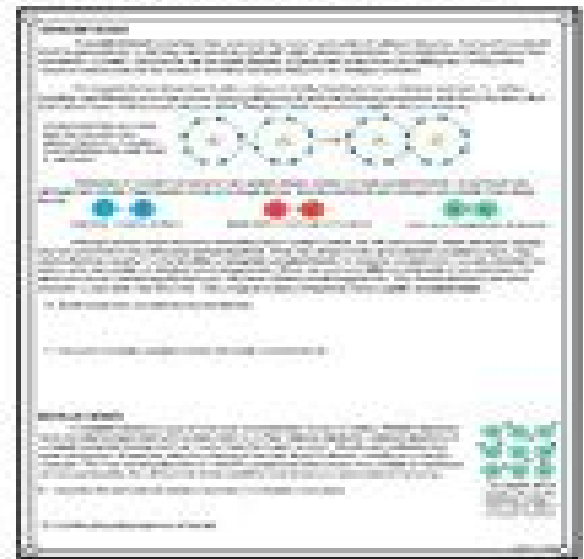
Properties of Ionic Compounds

1. High melting and boiling points due to the strong electrostatic forces between the ions.

2. Soluble in water, forming an aqueous solution of ions.

3. Conduct electricity when molten or in solution.

Diagram showing the structure of an ionic compound lattice.



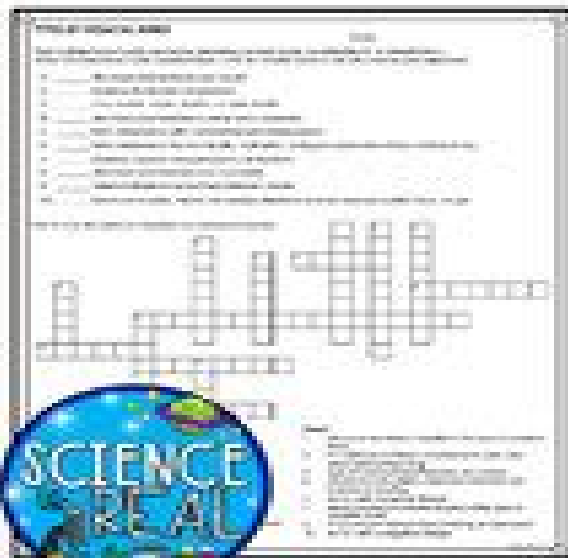
Properties of Covalent Compounds

1. Low melting and boiling points due to the weak intermolecular forces between the molecules.

2. Insoluble in water.

3. Do not conduct electricity.

Diagram showing the structure of a covalent molecule.



Properties of Metallic Compounds

1. High melting and boiling points due to the strong metallic bonds.

2. Soluble in water, forming an aqueous solution of ions.

3. Conduct electricity.

Diagram showing the structure of a metallic lattice.

- ✓ IONIC, COVALENT, & METALLIC BONDS
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Guided Reading 18 1 Chemistry

Daniela Niemeyer



Guided Reading 18 1 Chemistry:

Whispering the Techniques of Language: An Mental Journey through **Guided Reading 18 1 Chemistry**

In a digitally-driven world where displays reign supreme and quick interaction drowns out the subtleties of language, the profound strategies and emotional subtleties concealed within words frequently move unheard. However, situated within the pages of **Guided Reading 18 1 Chemistry** a charming fictional treasure pulsing with fresh emotions, lies a fantastic quest waiting to be undertaken. Published by a skilled wordsmith, this wonderful opus encourages visitors on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within ab muscles cloth of every word. Within the emotional depths of the touching evaluation, we will embark upon a sincere exploration of the book is primary styles, dissect their charming publishing model, and succumb to the strong resonance it evokes strong within the recesses of readers hearts.

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