

Quarter 3 Exam Review

Multiple Choice

Always use the correct state symbol when you write the balanced chemical equation.

- is a chemical reaction

 - the mass of the products is greater than the mass of the reactants
 - the mass of the product is greater than the mass of reactants
 - the material atoms in the reactants and products must change
 - energy is lost must be added to the reaction
- When a solid product is produced by a chemical reaction, sometimes the solution it is called

 - a precipitate
 - a solution
 - a reactant
 - the mass of the product
- In writing a chemical equation that produces hydrogen gas, the correct representation of hydrogen gas is

 - H
 - H₂
 - H₂g
 - H₂l
- To balance a chemical equation, it may be necessary to adjust the

 - coefficients
 - subscripts
 - charges of the products
 - number of products
- According to the law of conservation of mass, the total mass of the reacting substances is

 - always less than the total mass of the products
 - always less than the total mass of the products
 - sometimes more and sometimes less than the total mass of the products
 - always equal to the total mass of the products
- In the chemical equation $2\text{Mg}(s) + \text{O}_2(g) \rightarrow 2\text{MgO}(s)$

 - Mg represents the product magnesium
 - O₂ represents the reactant oxygen
 - Mg represents the reactant magnesium
 - O₂ represents the product oxygen gas
- When the reaction $2\text{FeCl}_3 + 3\text{H}_2\text{O} \rightarrow 2\text{Fe} + 6\text{HCl}$ is correctly balanced, what is the coefficient of Fe?

 - 4
 - 2
 - 6
 - 3
- Which coefficients correctly balance the formula equation $\text{CaCl}_2 + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$?

 - 1, 1, 2
 - 1, 2, 1
 - 1, 2, 1
 - 1, 1, 1
- In which kind of reaction do two or more substances combine to form a new compound?

 - decomposition reaction
 - single displacement reaction
 - combustion reaction
 - double displacement reaction
- The equation $\text{A}_2\text{X} \rightarrow \text{A} + \text{X}$ is the general equation for a

 - synthesis reaction
 - decomposition reaction
 - combination reaction
 - single displacement reaction

General Chemistry Review Exam Guide Answers Packet

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Table of Contents General Chemistry Review Exam Guide Answers Packet

1. Understanding the eBook General Chemistry Review Exam Guide Answers Packet
 - The Rise of Digital Reading General Chemistry Review Exam Guide Answers Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry Review Exam Guide Answers Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry Review Exam Guide Answers Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry Review Exam Guide Answers Packet
 - Personalized Recommendations
 - General Chemistry Review Exam Guide Answers Packet User Reviews and Ratings
 - General Chemistry Review Exam Guide Answers Packet and Bestseller Lists

5. Accessing General Chemistry Review Exam Guide Answers Packet Free and Paid eBooks
 - General Chemistry Review Exam Guide Answers Packet Public Domain eBooks
 - General Chemistry Review Exam Guide Answers Packet eBook Subscription Services
 - General Chemistry Review Exam Guide Answers Packet Budget-Friendly Options
6. Navigating General Chemistry Review Exam Guide Answers Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - General Chemistry Review Exam Guide Answers Packet Compatibility with Devices
 - General Chemistry Review Exam Guide Answers Packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry Review Exam Guide Answers Packet
 - Highlighting and Note-Taking General Chemistry Review Exam Guide Answers Packet
 - Interactive Elements General Chemistry Review Exam Guide Answers Packet
8. Staying Engaged with General Chemistry Review Exam Guide Answers Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry Review Exam Guide Answers Packet
9. Balancing eBooks and Physical Books General Chemistry Review Exam Guide Answers Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry Review Exam Guide Answers Packet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine General Chemistry Review Exam Guide Answers Packet
 - Setting Reading Goals General Chemistry Review Exam Guide Answers Packet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of General Chemistry Review Exam Guide Answers Packet
 - Fact-Checking eBook Content of General Chemistry Review Exam Guide Answers Packet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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