- According to the Bronsted—Lowry definition, which chemical species can function both as an acid and as a base?
- 17 a CI so,2

17

17

- c. NH.
- d. HCO₃
- e. H₃O*
- In the reaction CN⁻ + H₂O ←→ HCN + OH⁻ which is an acid—base conjugate pair?
 - a. H2O and HCN
 - A H₂O and OH
 - c. CN⁻ and H₂O
 - d. HCN and OH
- Given that HX is a stronger Bronsted acid than HY in aqueous solution, which is true of a 1 M solution of NaX
 - a. It is less basic than a 1 M solution of NaY.
 - It is more basic than a 1 M solution of NaY.
 - It yields a neutral solution.
 - It is more concentrated than a 1 M solution of NaY.
- 4. The pH of a 0.03 M HCl solution is
- a. 1.5
 - b. 2.5
 - c. 3.5 d. 12.5
 - 5. The pH of a 1.0 x 10⁻³ M Ba(OH)₃ solution at
 - 25°C is a. 2.7
 - a. 2.7 A. 3.0
 - c. 11.0
 - d. 11.3
 - 6. Which is the strongest acid?
- a. HCiO
 - & HClO₂
 - c. HClO₃
 - d. HClO₂
 - 7. Which salt reacts with water (hydrolyzes) to produce a basic solution?
- F & NaC2H3O2
 - A. NaNO
 - c. NH₄Cl
 - at BaSO.

8. The weakest of the bases listed is

Aeid	Conjugate Base	Ka (Ionization Constant of Acid)	
HCI	CI	100% ionized	
HSO ₄	SO,2	1.2 x 10 ⁻²	
H ₂ S	HS	5.7x10 ⁻⁸	
HIS	S	1.2×10 ⁻¹⁵	7

at CE

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- & CN
- c. HS
- d. 52-
- e. SO₂2-

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Table 1-1	100			-			7	-	



- Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate KC₂H₃O₂, is less alkaline than a 0.10 M solution of potassium cyanide, KCN?
- A. Hydrocyanic acid is a weaker acid than acetic acid.
 - Hydrocyanic acid is less soluble in water than acetic acid.
 - Cyanides are less soluble than acetates
 - A Acetic soid is a weaker soid than hydrocyanic soid
 - In the titration of 50.0 mL of 0.100 M benzoic acid (a monoprotic acid) with 50.0 mL of 0.100 M NaOH, the properties of the solution at the equivalence point will correspond exactly to the properties of
 - a. a 0.100 M sodium solution.
 - a 0.0500 M sodium hydroxide solution.
 - c. a 0.0500 M benzoic acid solution.
 - d. a 0.0500 M sodium benzoate solution.

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Mastery Through Quizzing Stan Skrabut, 2025-03-24 Clear your desks You have a quiz Few phrases spark more anxiety in students The tension the stomach knots the panicked glances around the room We ve all been there But what if quizzing wasn t something to dread What if it became the key to deeper learning greater confidence and real mastery instead Mastery Through Quizzing isn t just about testing knowledge but transforming how we learn Whether you re an educator administrator or instructional designer this book will show you how to turn low stakes assessments into powerful tools for engagement retention and long term success Discover how to Shift from high pressure exams to a growth focused quizzing strategy that reduces anxiety and boosts confidence Design questions that drive critical thinking not just memorization Leverage technology and test banks to make quizzes more effective and scalable Implement a step by step Mastery Quizzing Strategy to help students achieve real understanding Unlike traditional assessments that measure what students don t know mastery quizzing helps them build knowledge step by step reinforcing what they know and guiding them forward With decades of experience in instructional technology and education I ve seen firsthand how this method can transform classrooms I m sharing the approach to help you move beyond outdated testing models and create a learning experience that works It s time to rethink quizzing Let s turn it from a source of stress into a stepping stone for mastery Scroll up and grab

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