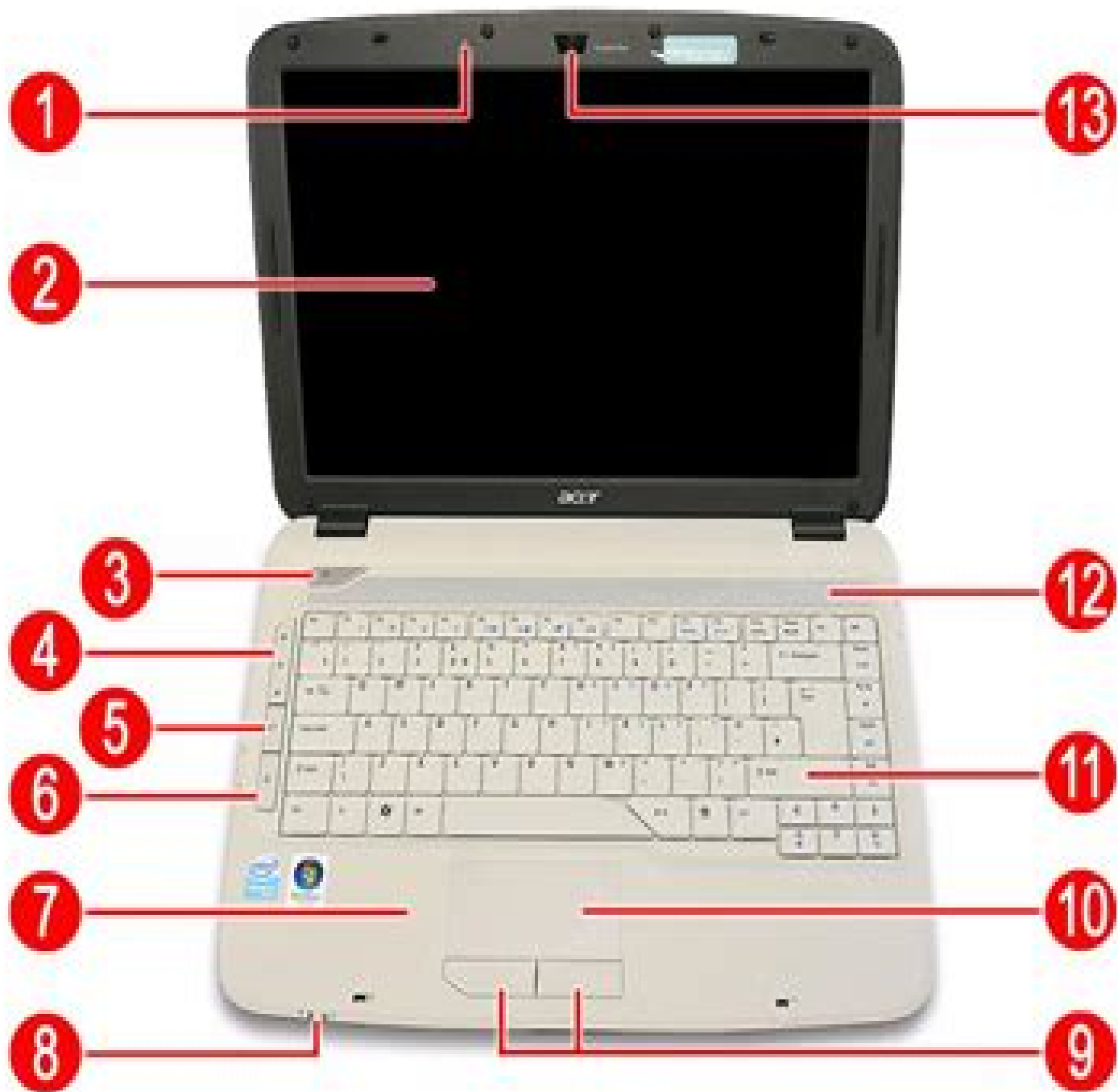




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web in a chemical equilibrium the forward and reverse reactions occur at equal rates and the concentrations of products and reactants remain constant law of mass action the law of mass action links the rate of a chemical reaction as proportional to the concentrations of the reactants and products in a chemical reaction for a chemical reaction

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web equilibrium the rate of the forward reaction and the rate of the reverse reaction are equal at equilibrium both reactants

and products are present some reaction mixtures contain mostly reactants and form only a few products at equilibrium some reaction mixtures contain mostly products and remain mostly reactants at equilibrium

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web chemical equilibrium part 1 forward and reverse reactions understanding the concept of chemical equilibrium is critical to following several of the discussions that we have in bis2a and indeed throughout biology and the sciences