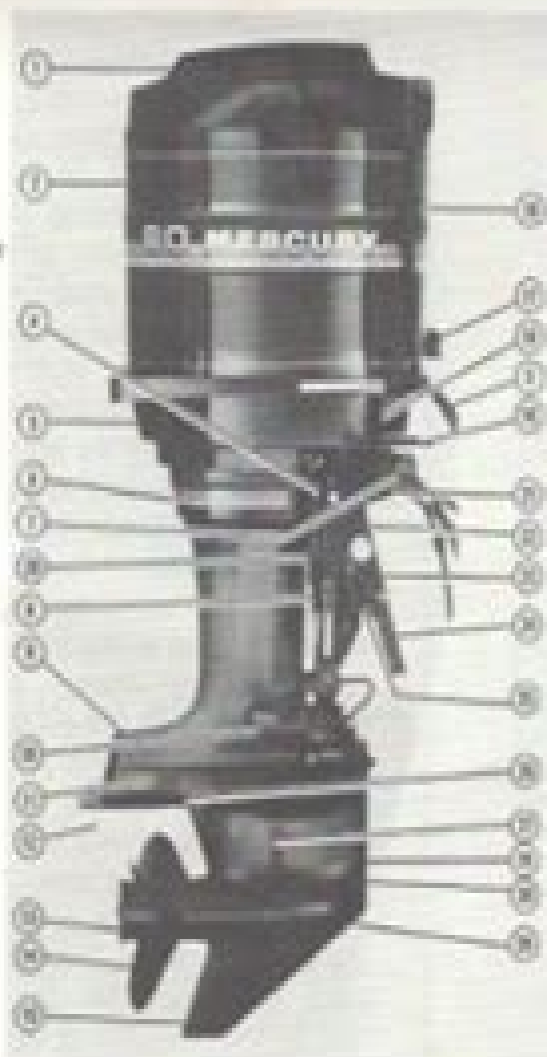


1 90 - Typical

1. Top Cowl
2. Wrap-Around Cowl
3. Electrical Harness
4. Tilt Stop Lever
5. Water Pump 'Tall-Tale' Hole
6. Trim Cover
7. Tilt Tube
8. Dyna-Shock Absorber
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13. Jet-Prop Exhaust
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15. Flap
16. Front Cowl
17. Fastening Knob
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19. Steering Arm
20. Serial Number Plate
21. Clamp Bracket (2)
22. Upper Transom Bolt Hole
23. Clamp and Swivel Bracket
24. Lower Transom Bolt Hole
25. Tilt Lock Pin
26. Anti-Cavitation Plate
27. Water Intake
28. Gear Housing
29. Oil Fill and Drain
30. Speedometer Pickup



This owner's publication includes operation and service instructions. If disassembly or replacement, particularly of internal parts, is required, the owner is advised to see an Authorized Service Dealer and not to attempt the repair work himself.

SPECIFICATIONS

	140 HP	115 HP	90 HP	90 HP
Bore	73mm (2.78")			
Stroke	65mm (2.56")			
Total Piston Displacement	1635cc (99.8 Cu. In.)			1091cc (66.6 Cu. In.)
Recommended Spark Plugs	Champion QL 7EV			Champion QL 7EV
Rated Horsepower (Continuous Duty)	99 kw (134 HP)	81 kw (110 HP)	64 kw (86 HP)	57 kw (76 HP)
Full Throttle RPM Range	5300-5800	5000-5500	4600-5200	5000-5500
Fuel Tank Capacity	22.5 Liters (5 Imperial Gallons; 6 U.S. Gallons)			

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DJ Losen



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web nov 20 2020 in this video we ll balance the equation $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{H}_2 + \text{H}_2\text{O}$ and provide the correct coefficients for each compound to balance $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{H}_2 + \text{H}_2\text{O}$ you ll need to be

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web magnesium hydroxide nitric acid balanced total equation $\text{Mg(OH)}_2 + 2\text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + 2\text{H}_2\text{O}$ aq $\text{Mg(NO}_3)_2$ aq total ionic equation $\text{Mg}^{2+} + 2\text{OH}^- + 2\text{H}^+ + 2\text{NO}_3^- \rightarrow \text{Mg}^{2+} + 2\text{NO}_3^- + 2\text{H}_2\text{O}$ net ionic equation $2\text{OH}^- + 2\text{H}^+ \rightarrow 2\text{H}_2\text{O}$ sulfuric acid and lithium hydroxide balanced total equation $\text{H}_2\text{SO}_4 + 2\text{LiOH} \rightarrow \text{Li}_2\text{SO}_4 + 2\text{H}_2\text{O}$

