

DSC-S80

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The Cyber-shot S80 has a traditional camera feel and features a large 2" LCD screen, 4.1 effective Megapixels and a high-quality Carl Zeiss Vario-Tessar 3x optical zoom lens.



Features

Please be aware that the features/specifications can differ from country to country.

- Large 2.0" LCD
- 4.1 Effective Megapixel
- 3x Optical Zoom, up to 10x Smart Zoom, 6x Precision Digital Zoom
- Carl Zeiss® Vario-Tessar® Lens
- Stamina Battery Life (approx 440 shots by CIPA standard with supplied battery NH-AA-2DA)
- Rechargeable Batteries & Charger included
- Quick Operation (Shutter Time Lag 0.3 sec, Shutter Release Lag 0.009 sec)
- Real Imaging Processor
- Manual Exposure Setting
- MPEG iVX Fine with Audio
- AF Illuminator
- Histogram Indication
- USB 2.0 High Speed Transfer
- Scene Selection Mode
- Internal Memory 32MB
- 100% LCD Field of view
- PictBridge Compatible
- Conversion Lens Compatible (Optional)

Technical Specifications

This document was created on
19-Jul-2006

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Manual Cyber Shot Dsc S8digital Camera

IM Harris



Manual Cyber Shot Dsc S8digital Camera:

Unveiling the Magic of Words: A Report on "**Manual Cyber Shot Dsc S8digital Camera**"

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{ag}^+\text{aq} + \text{cr}_2\text{O}_7^{2-}\text{aq} \rightarrow \text{ag}_2\text{cr}_2\text{O}_7\text{s}$ label 4 2 3

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web mar 11 2014 solution ionic equation $\text{h}^+\text{aq} + \text{cl}^-\text{aq} + \text{na}^+\text{aq} + \text{hco}_3^-\text{aq} \rightarrow \text{h}_2\text{O} + \text{co}_2\text{g}$ the spectator ions are cl⁻ and na⁺ net ionic equation $\text{h}^+\text{aq} + \text{hco}_3^-\text{aq} \rightarrow \text{h}_2\text{O} + \text{co}_2\text{g}$ h aq hco₃ aq h₂o l co₂ g here is a video on spectator ions *chapter 8 04 ionic equations chemistry libretexts* - Mar 06 2023

web the six no₃⁻ aq ions and the six na⁺ aq ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{ba}^{2+}\text{aq} + 2\text{po}_4^{3-}\text{aq} \rightarrow \text{ba}_3\text{po}_4\text{s}$

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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 $\text{AgNO}_3(\text{aq}) + \text{KCl}(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{KNO}_3(\text{aq})$ total ionic $\text{Ag}^+(\text{aq}) + \text{NO}_3^-(\text{aq}) + \text{K}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{K}^+(\text{aq}) + \text{NO}_3^-(\text{aq})$ net ionic $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s})$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l})$ the spectator ions in $\text{KOH}(\text{aq})$ and $\text{HNO}_3(\text{aq})$ are $\text{K}^+(\text{aq})$ and $\text{NO}_3^-(\text{aq})$ how to identify and list the

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web the aq aq and cl aq ions become agcl s but the na aq ions and the no 3 aq ions stay as na aq ions and no 3 aq ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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