

Pass the Proton Acid-Base Theories

Name _____

Date _____ Period _____

Purpose

To compare acid and base solutions at the particulate level.

Materials

- Acid-Base Solution cards

Procedure and Questions

- Sort the cards into three groups based on common features. Write the solution card number, molarity, chemical formula, and chemical name for each solution.

Group 1

4: CHOOH
1: HCL
5: CH₂COOH
6: NaCl
9: NaOH
7: CH₂OH
11: NH₄

All have 0.010 M

Group 2

2: HCL
8: CH₂OH

All have 0.005 M

Group 3

3: HCL
10: NaOH
12: HOH

All have 0.001 M

- What do the solutions in each of the three groups have in common?

Group 1: same molarity, 0.010 M and a lot of protons

Group 2: same molarity, 0.005M and little protons

Group 3: same molarity, 0.001 M and lack of protons

- Is there a second way you might sort the cards? If so, explain how.

You can sort them by the patterns/shape of protons

- Fill in the table with the chemical formula of the substance dissolved and the ions in the solution.

Lesson 12 Unit 4 Toxins Worksheet

VM Jensen

A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the VM Jensen text.

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