HSC Maharashtra Board question paper: March 2013

Note:

- All question are compulsory
- ii. Answer to the two sections are to be written in the same answer book.
- iii. Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Encery new question must be started on a new page.
- vi. Use of logarithmic table is allowed

CHEMISTRY: SECTION - I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7

i. Ir	n body centred cubic structure the space occupied is about				
(2	(A)	68 %		(B)	53 %
(0	C).	38 %		(D)	32 %
ii. Fe	For a gaseous reaction the unit of rate of reaction is				
(2	A)	L atm s ⁻¹		(B)	arm mol ⁻¹ s ⁻¹
(6	C)	atm s 1		(D)	mol s
iii. W	Which of the following compounds contains S = O as well as S = S bonds?				
C2	(A)	Sulphuric acid		(B)	Thiosulphuric acid
(6	C)	Sulphurous acid	00	(D)	Thiosulphurous acid
iv. W	Which of the following solutions shows maximum depression in freezing point?				
(4	(A)	0.5 M Li ₂ SO ₄		(B)	1 M NaCl
0	C)	0.5 M Al ₂ (SO ₄) ₅) "	(D)	0.5 M BaCl ₂
v. F	For a chemical reaction $\Delta S = -0.035$ kJ/K and $\Delta H = -20$ kJ.				
(4	A)	5.14 K		(B)	57.14 K
6	2)	571.4 K		(D)	5714.0 K
vi. T	The standard e.m.f of the following cell is 0.463 V				
C	Su	Cum Agan Ag. I	If Eas - 0.800 V,		

vii. Fe₂O₃ is reduced to spongy iron near the top of blast furance by

(A) H₂

(A) 1.137 V

(C) 0.463 V

(B) CaO

0.337 V

-0.463 V

(B)

(ID)

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What is the standard potential of Cu electrode?

Hsc Questions 2013

Burton St. John III, Yvette E. Pearson

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